

# Visibility Overview

Jenny Hand

CIRA, Colorado State University

1990

San Geronio, CA

2023



# We'll talk about:

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1. What is visibility?
2. What causes haze?
3. What is light extinction?
4. Visibility measurements
5. 2<sup>nd</sup> IMPROVE Equation
6. Other visibility parameters (visual range, deciview)



# 1. What is Visibility?

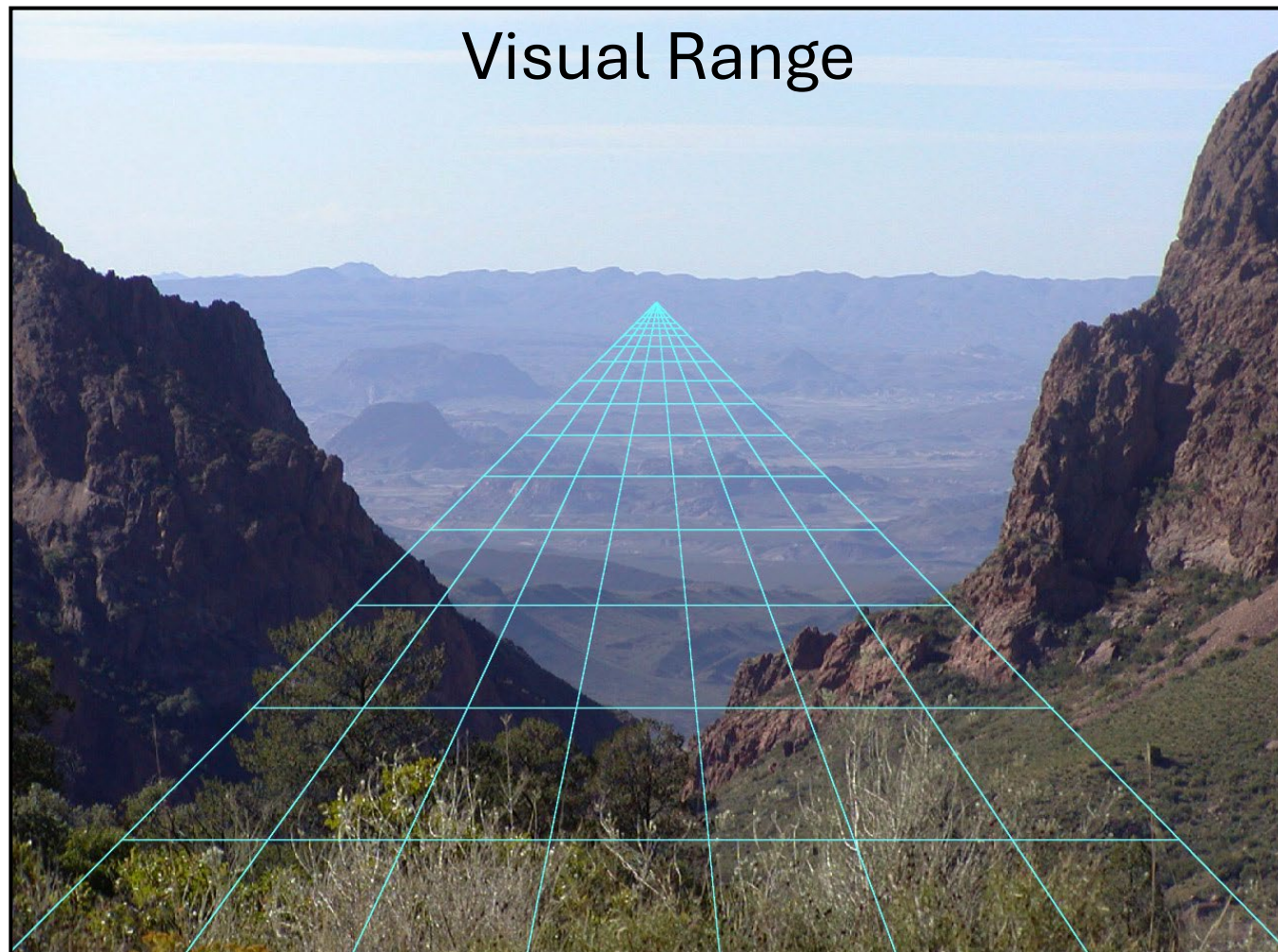
**Visibility is historically defined as visual range:**

*“the farthest distance one can see a large black object against the background sky.”*

-Useful for aviation, doesn't account for changes in scene, such as loss of texture or discoloration as haze increases.

**Visibility is more than just how far we can see.**

-It is better described as how “well” we can see and appreciate the colors, textures, forms, and detail in distant landscape features.



Theoretical limit of visual range is 243 miles

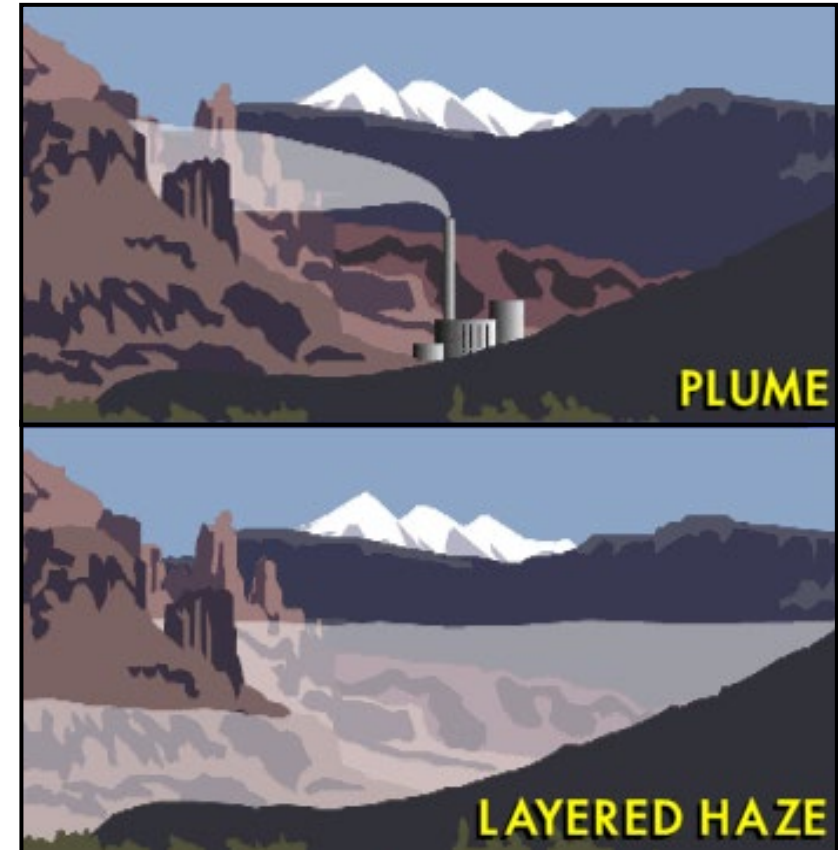
# Types of Haze

How impairment manifests depends upon the extent and distribution of particles and gases in the atmosphere.



**Visibility Impairment:** generally associated with discoloration, haziness, and loss of color and detail.

**Uniform Haze:** Pollutants are uniformly distributed from the ground to a height well above the highest terrain feature.



**Plume:** Pollutants are constrained in a tight elevated layer that can often be traced to a nearby source.

**Layered Haze:** Pollutants are often trapped near the ground beneath a temperature inversion. The top edge of the pollutant layer is visible.



# Examples of different types of haze layers



Grand Canyon National Park, Arizona



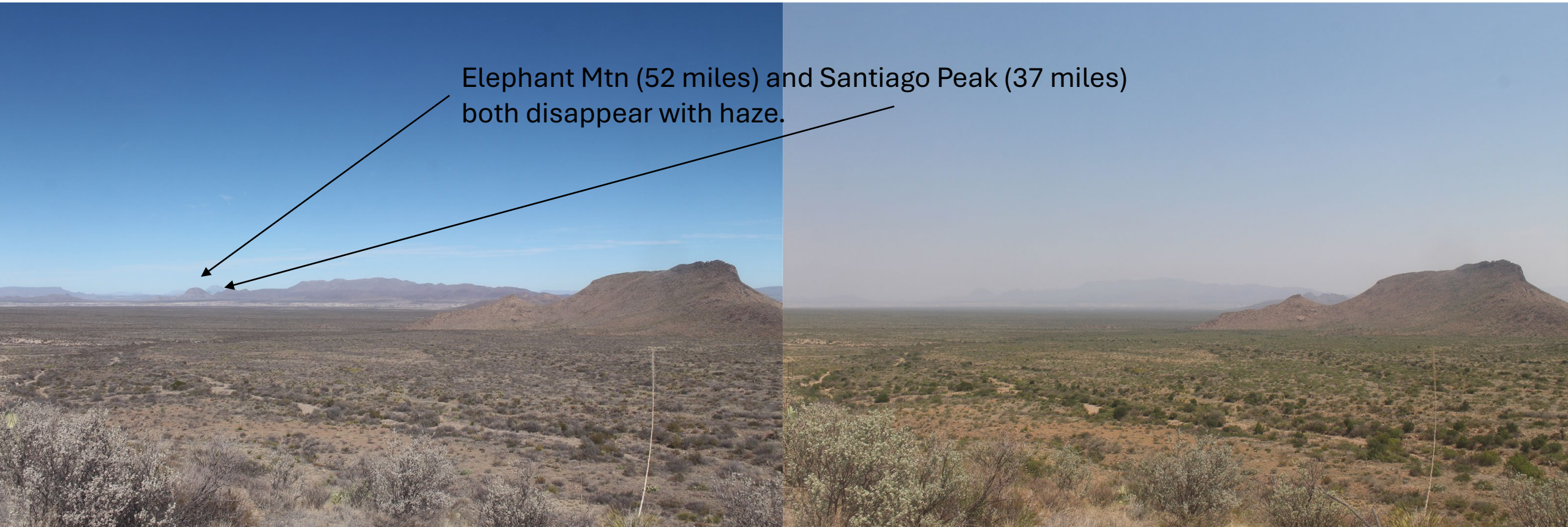
# Great Smoky Mountains National Park, TN

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# Big Bend National Park, TX



Elephant Mtn (52 miles) and Santiago Peak (37 miles)  
both disappear with haze.



# Sequoia & Kings Canyon National Park, CA

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# Sequoia & Kings Canyon National Park, CA

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# Dinosaur National Monument, CO

Ammonium  
nitrate





# Anthropogenic and Natural Sources of Haze

El Paso, TX; March, 2012  
( $PM_{10} > 5000 \mu\text{g m}^{-3}$ )

Credit: Joseph Rogash, NOAA-NWS,  
courtesy of Tom Gill, UTEP

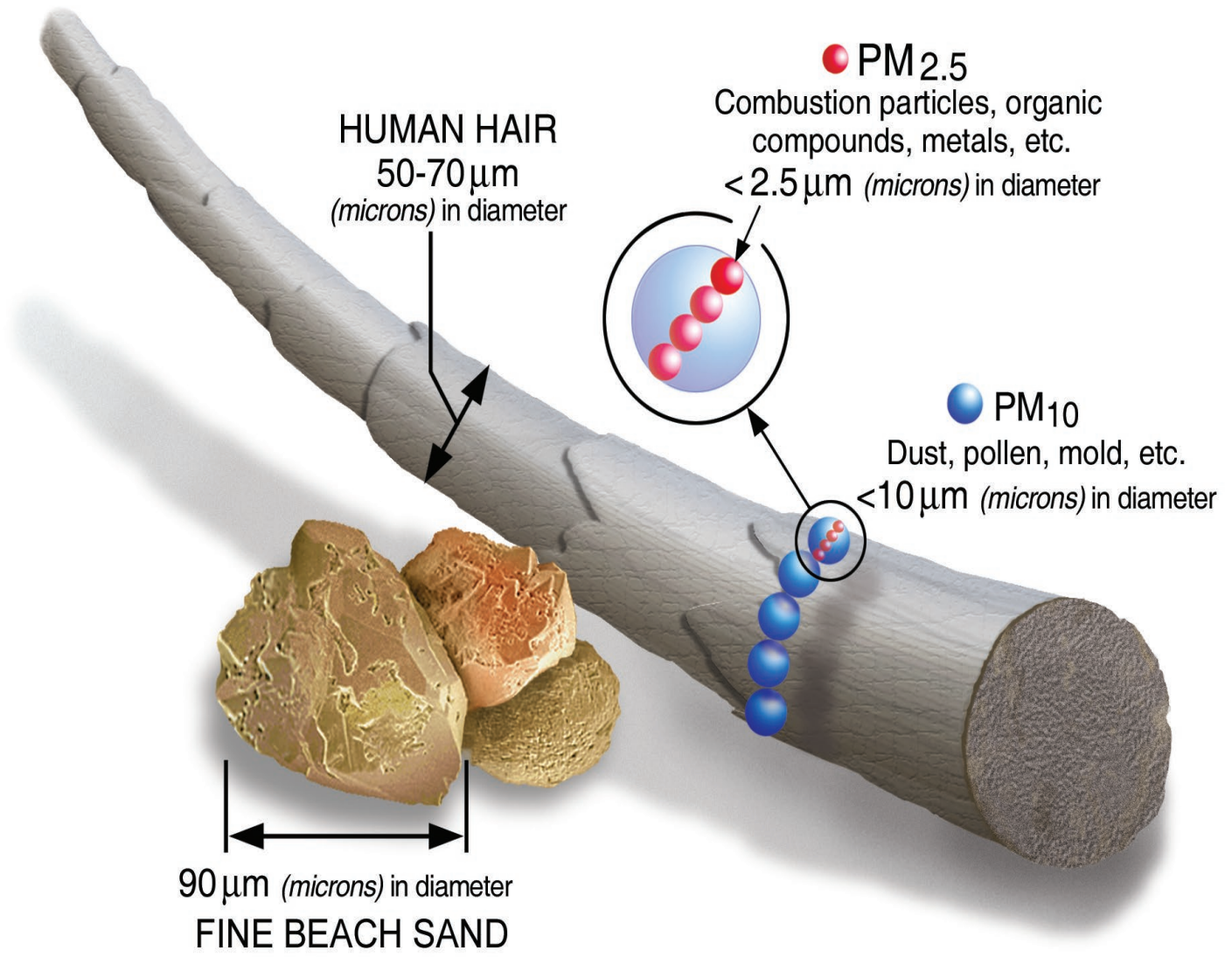
Williston Basin, ND

Credit: Bill Malm

Fort Collins, CO  
10/14/2020  
fcgov.com

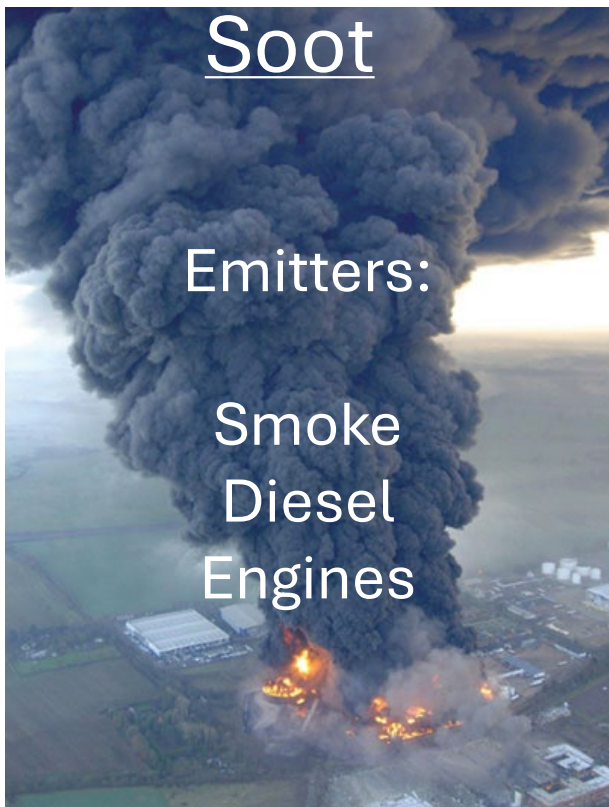


# 2. Causes of Haze: Atmospheric Aerosols





# Sources of Aerosols: Natural and Anthropogenic



Soot

Emitters:


- Smoke
- Diesel Engines



Dust

Emitters:

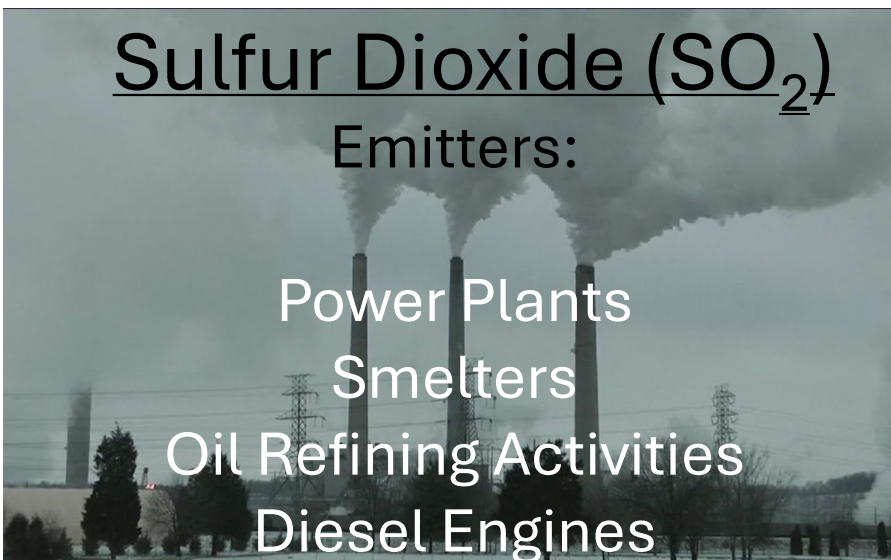
- Drylands
- Agricultural Activities
- Fugitive Dust



Hydrocarbons

Emitters:

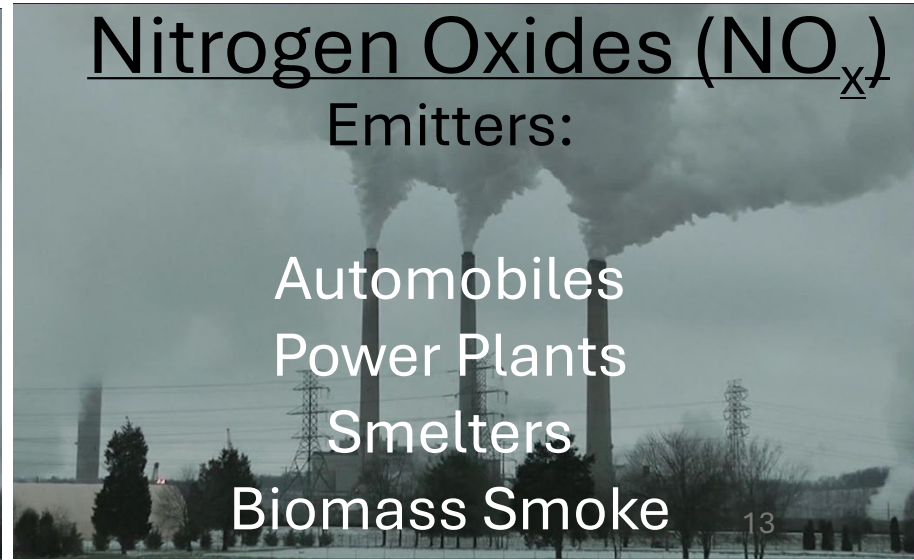
- Gas:
- Trees
- Automobile Exhaust
- Industrial Emissions
- Cooking
- Diesel Engines
- Direct:
- Biomass Burning



Sulfur Dioxide (SO<sub>2</sub>)

Emitters:

- Power Plants
- Smelters
- Oil Refining Activities
- Diesel Engines

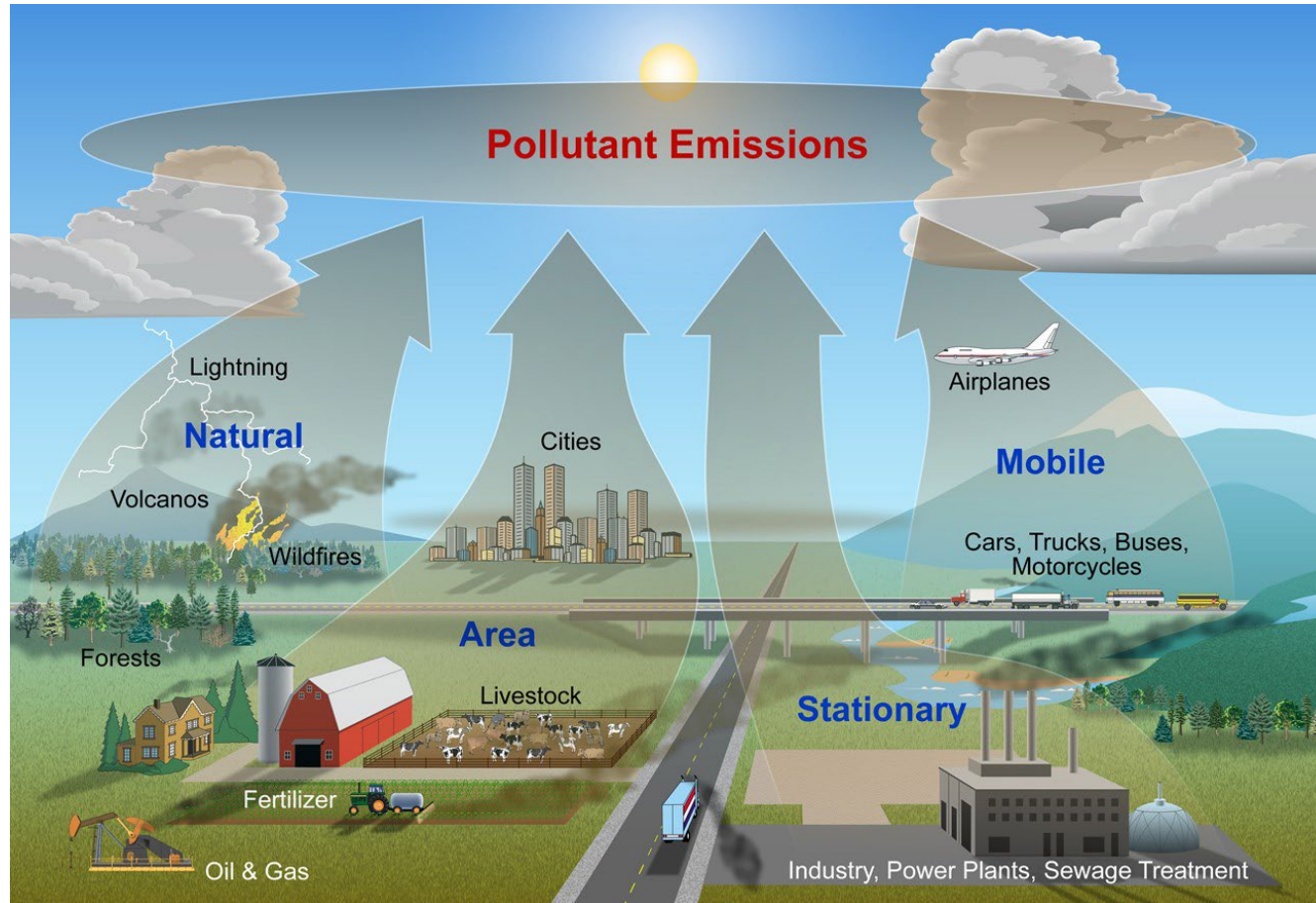


Nitrogen Oxides (NO<sub>x</sub>)

Emitters:

- Automobiles
- Power Plants
- Smelters
- Biomass Smoke

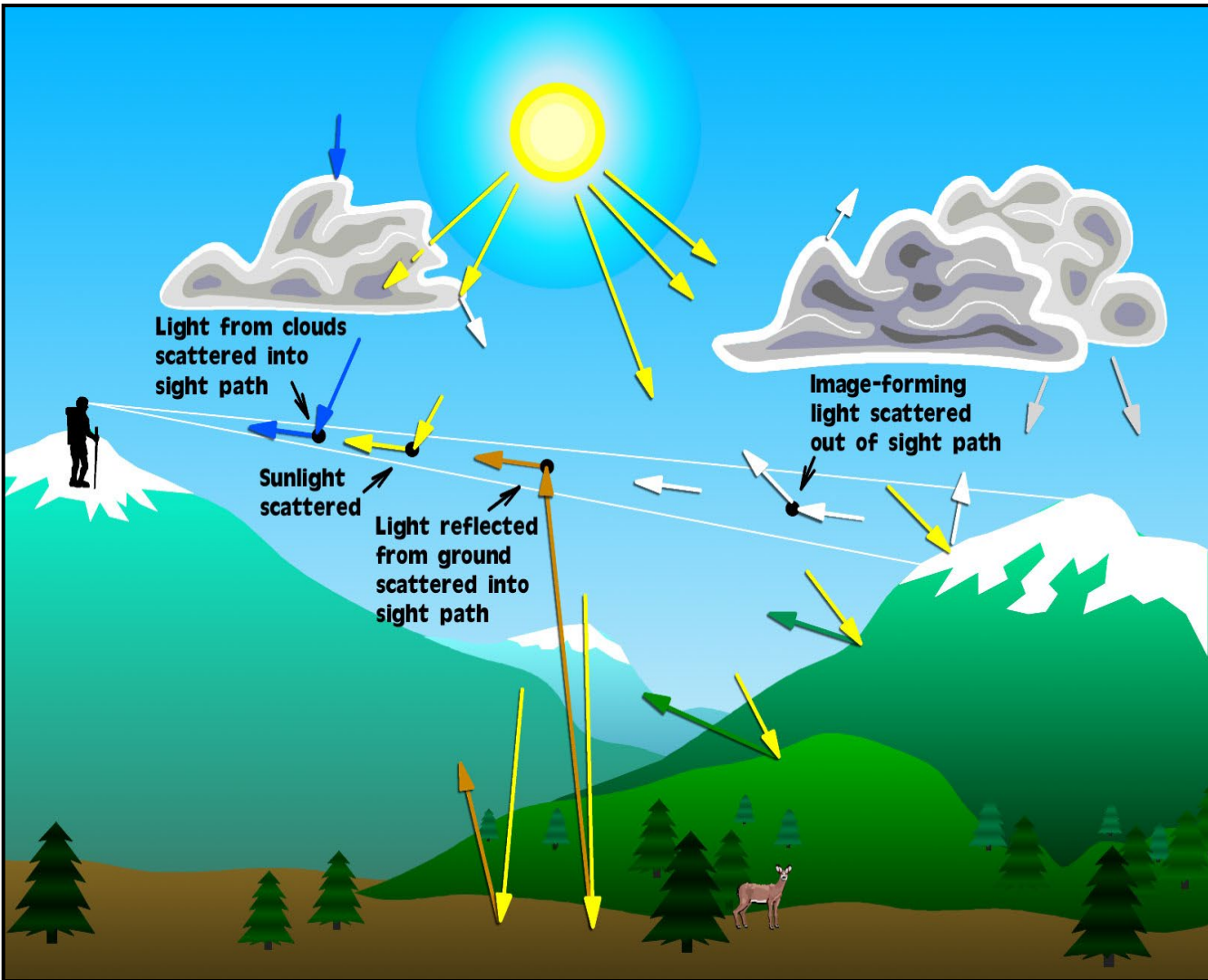
# Secondary Particle Formation



Precursor emissions disperse in the atmosphere, convert into secondary particles through complex atmospheric chemical reactions, then travel long distances to deposit in remote areas far from their source.

# How Pollutants Cause Haze

## Scattering in the Atmosphere



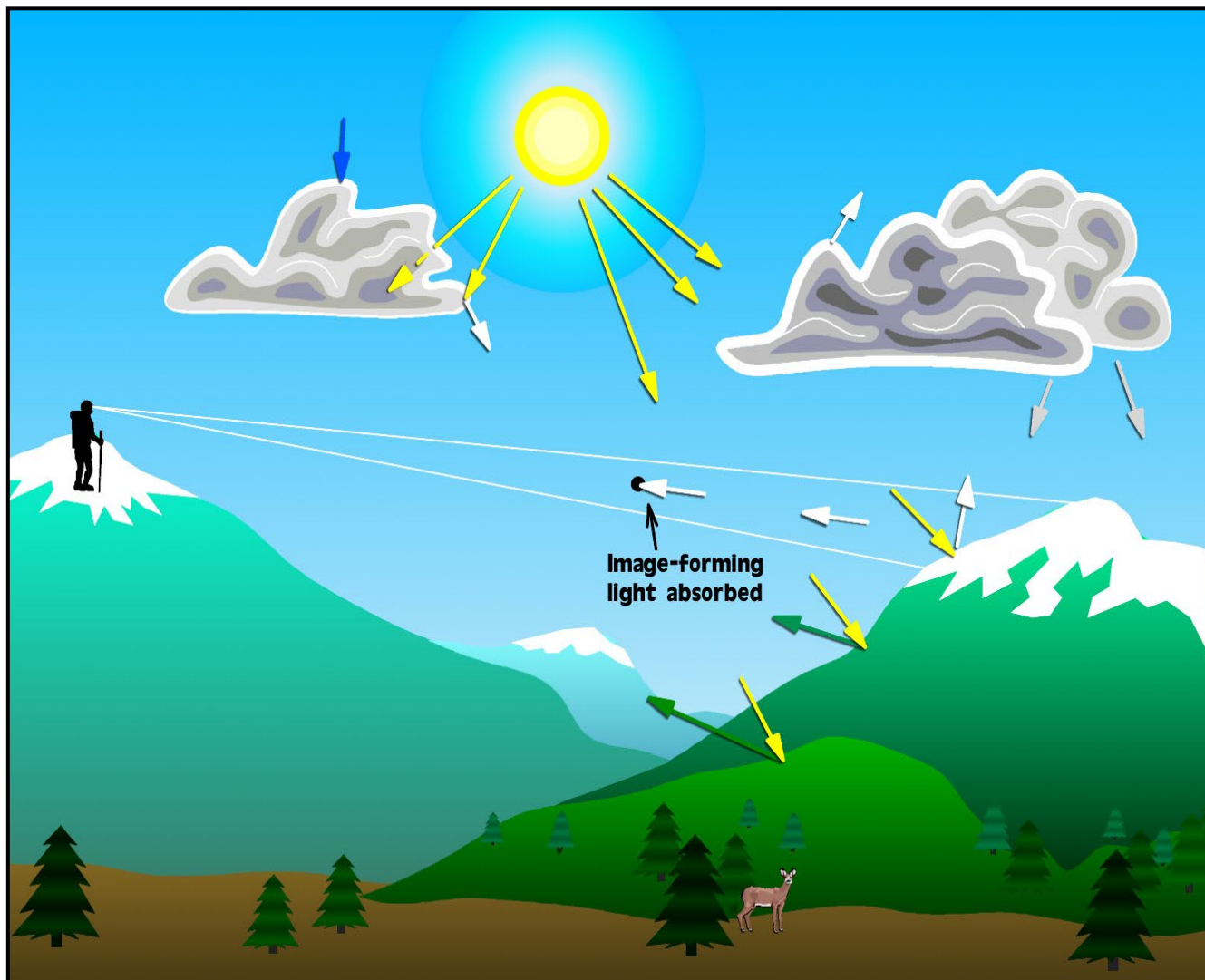
Particles and gases in the atmosphere can scatter or redirect image-forming light as it travels to the eye.

Through scattering, some image-forming light is removed from the view path.

In addition, extra light, sunlight, and light reflected from the clouds and ground are added to the sight path, which interferes with the ability to view the scene.

# How Pollutants Cause Haze

## Absorption in the Atmosphere



Another cause of visibility impairment is absorption.

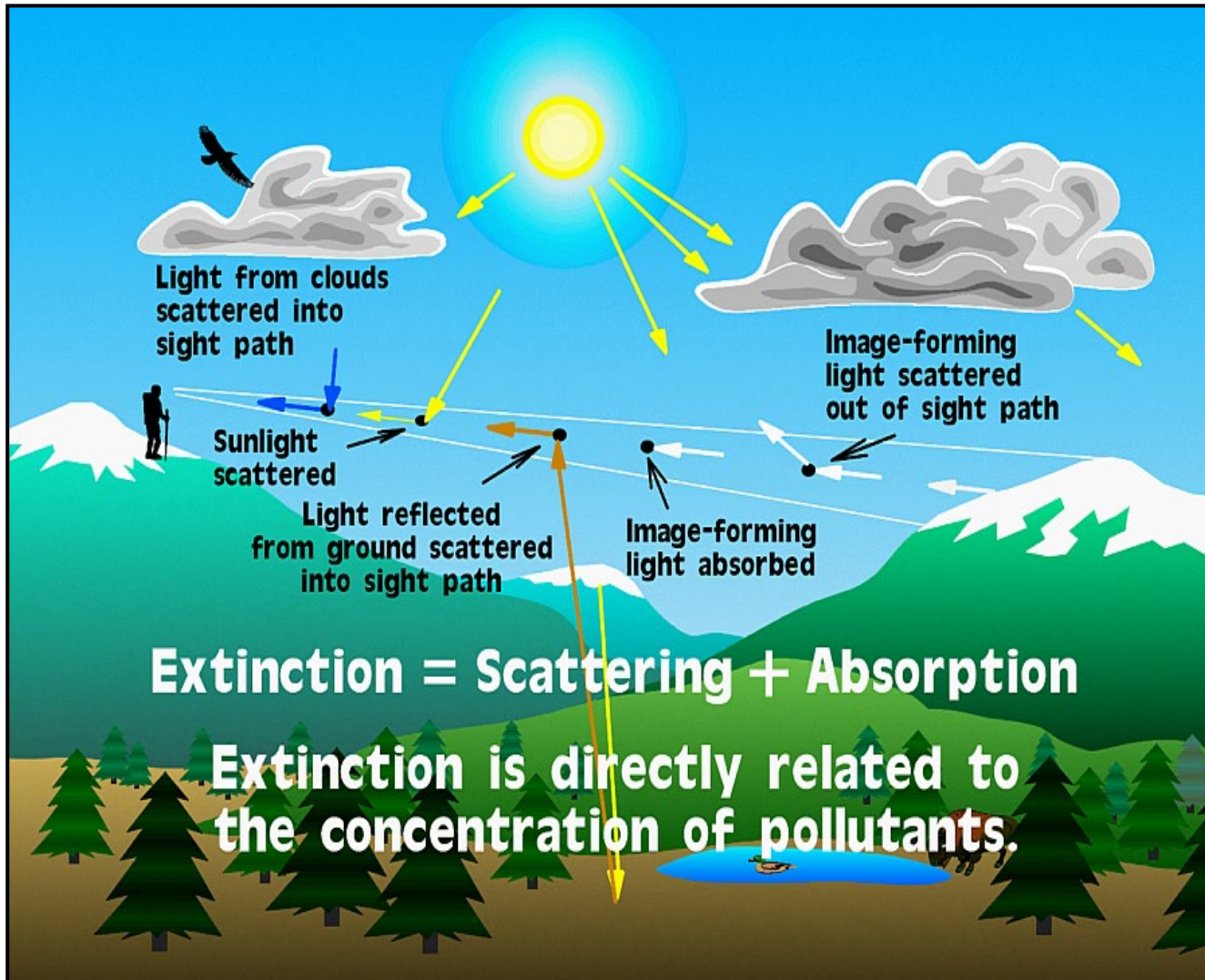
Particles and gases in the atmosphere absorb or remove image-forming light before it ever reaches the viewer's eye.

Although significant, absorption usually is less important than scattering processes when we talk about visibility impairment.



# How Pollutants Cause Haze

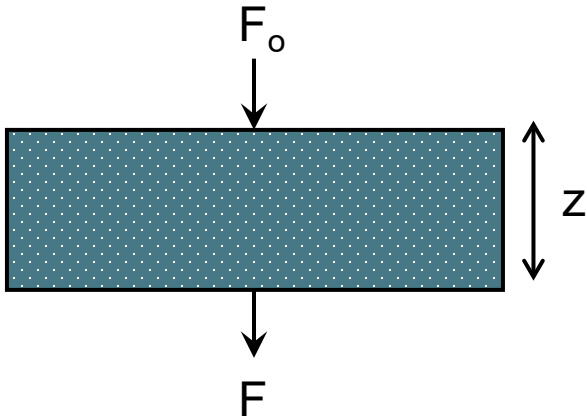
## Extinction in the Atmosphere



Extinction is a visibility metric used to describe the combined effect of scattering and absorption. It is proportional to the total amount of light removed as light passes through the atmosphere and is related to the concentration of pollutants.

# 3. Light Extinction Coefficients

Attenuation of incident light by scattering and absorption as it passes through a layer is described by Beer's law:



$$\frac{F}{F_o} = \exp(-b_{ext}z)$$

$F_o$  = incident light

$F$  = transmitted light

$b_{ext}$  is the extinction coefficient ( $Mm^{-1}$ )

$z$  is path length

$$b_{ext} = b_{sp} + b_{ap} + b_{sg} + b_{ag}$$

$b_{sp}$ ,  $b_{ap}$ : Scattering and absorption by particles

$b_{sg}$ ,  $b_{ag}$ : Scattering and absorption by gases

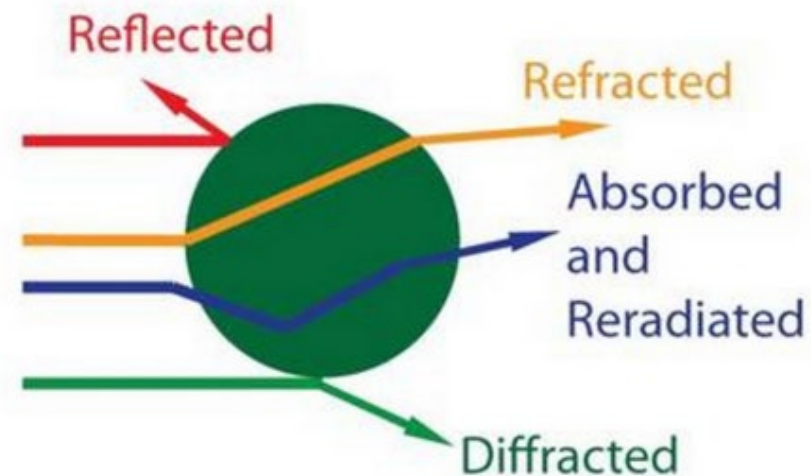
Rayleigh scattering by molecules,  $NO_2$  absorption in visible wavelengths



# Aerosol Radiative Effects

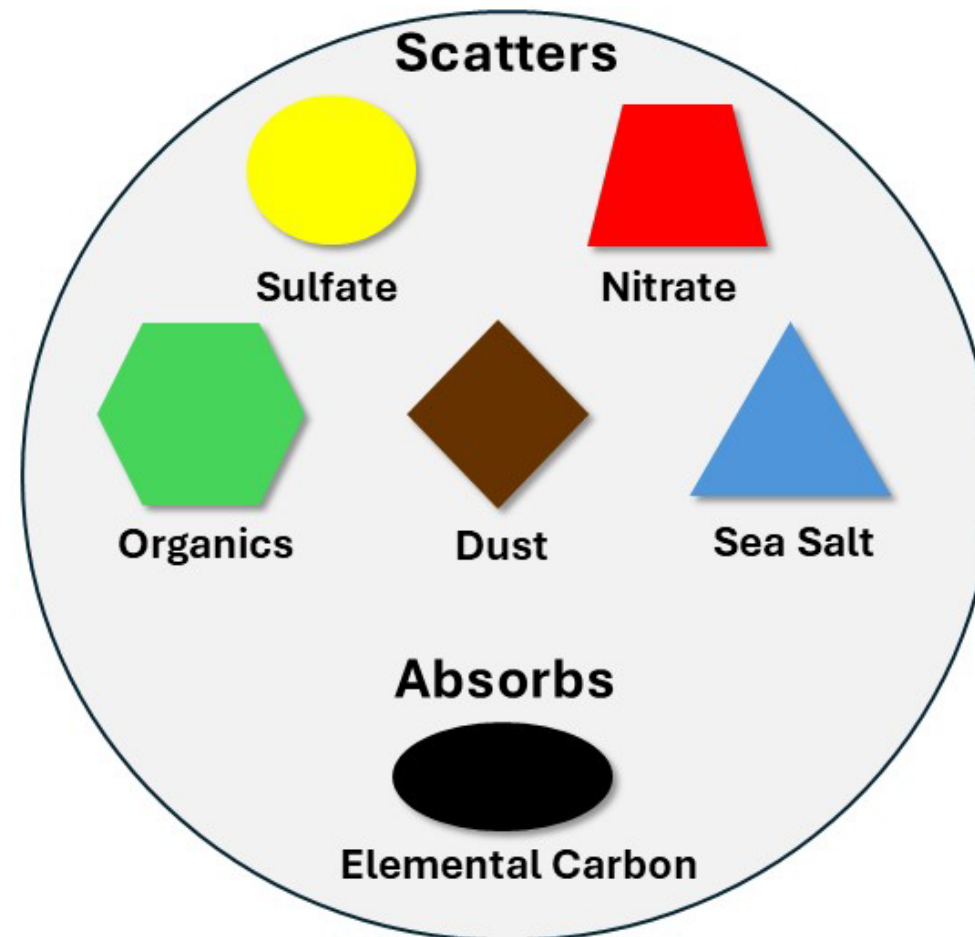
Aerosol interaction with solar radiation depends on:

- particle composition (scattering/absorbing)
- particle size
- ability to absorb water (hygroscopicity)
- particle shape



# Aerosol Composition

Degree of scattering (or absorption)  
depends on optical properties  
(refractive index)



# Aerosol Size

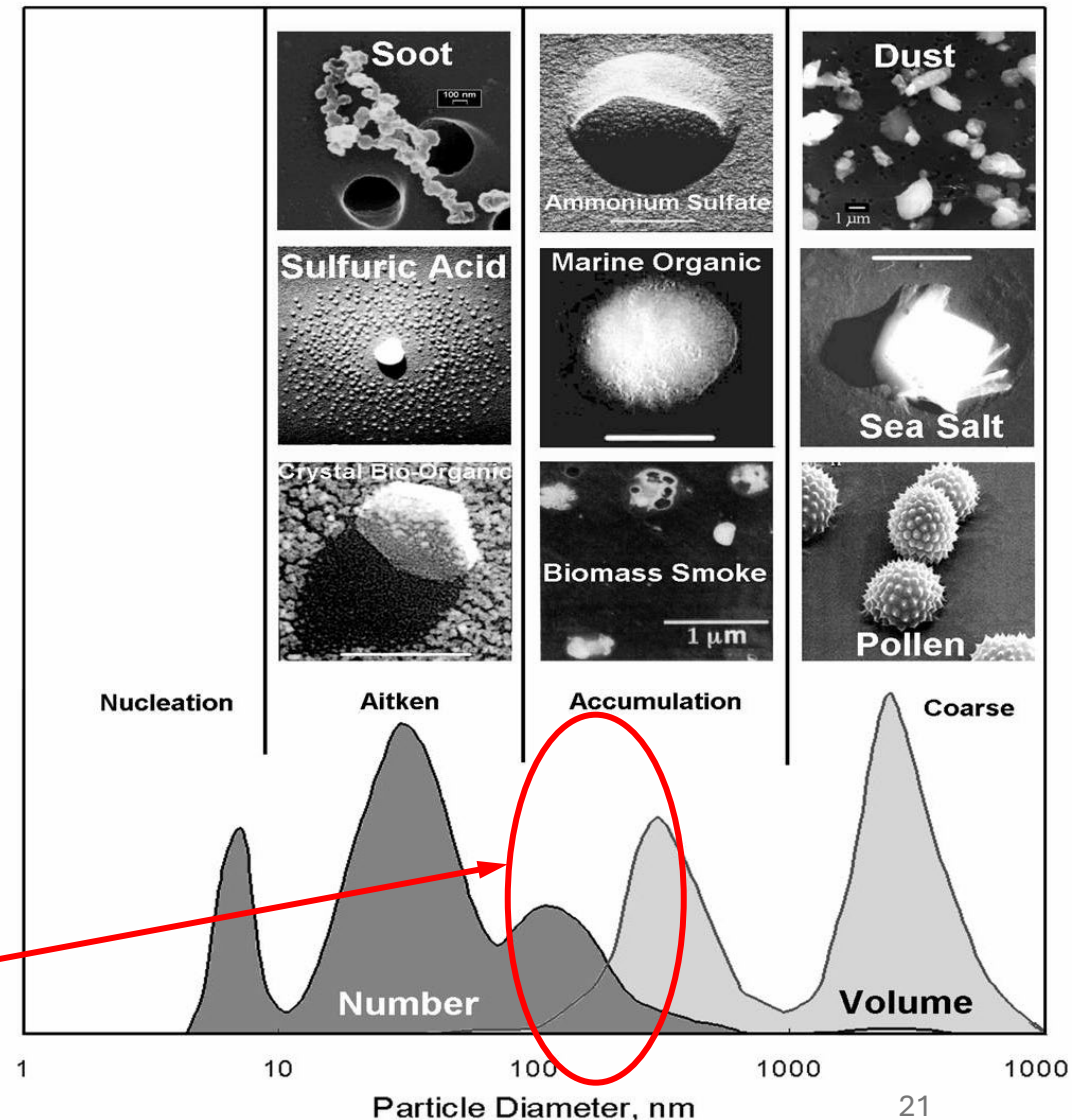
## PM<sub>2.5</sub> (Fine Particles):

- Particles with aerodynamic diameters less than 2.5 μm.
- These particles can stay suspended for weeks and are transported far from their source.

## PM<sub>10</sub> (Coarse Particles):

- Particles with aerodynamic diameters less than 10 μm.
- These particles usually deposit out of the air close to their source.

This size range interacts most efficiently with solar radiation





# Particle Size Effects on Haze

**Base Case:**



1  $\mu\text{g m}^{-3}$  Ammonium Sulfate  
1  $\mu\text{g m}^{-3}$  Coarse Mass  
 $b_{\text{ext}} = 13 \text{ Mm}^{-1}$

**Coarse mass:**

Increase has less effect



1  $\mu\text{g m}^{-3}$  Ammonium Sulfate  
**20  $\mu\text{g m}^{-3}$  Coarse Mass**  
 $b_{\text{ext}} = 24 \text{ Mm}^{-1}$

**Fine mass:**

Increase has bigger effect

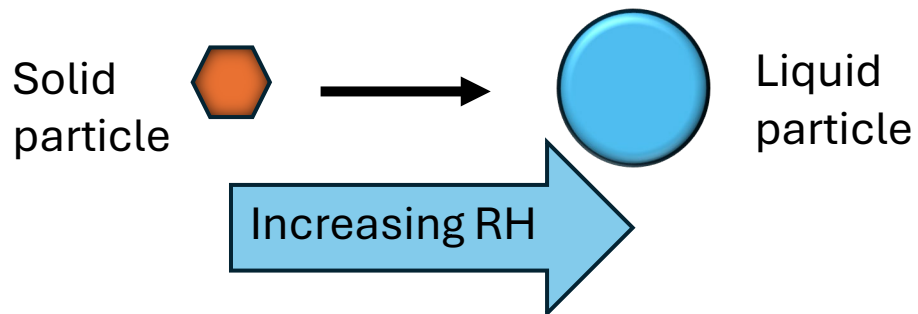


**20  $\mu\text{g m}^{-3}$  Ammonium Sulfate**  
1  $\mu\text{g m}^{-3}$  Coarse Mass  
 $b_{\text{ext}} = 107 \text{ Mm}^{-1}$

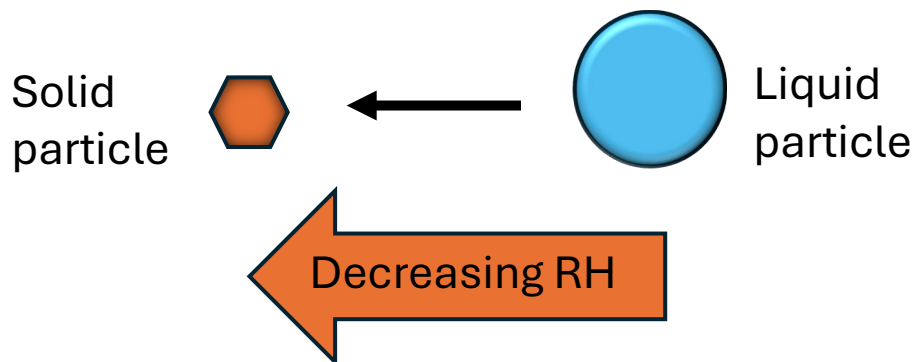


# Aerosol Growth Curves (Hygroscopicity, $D/D_0$ )

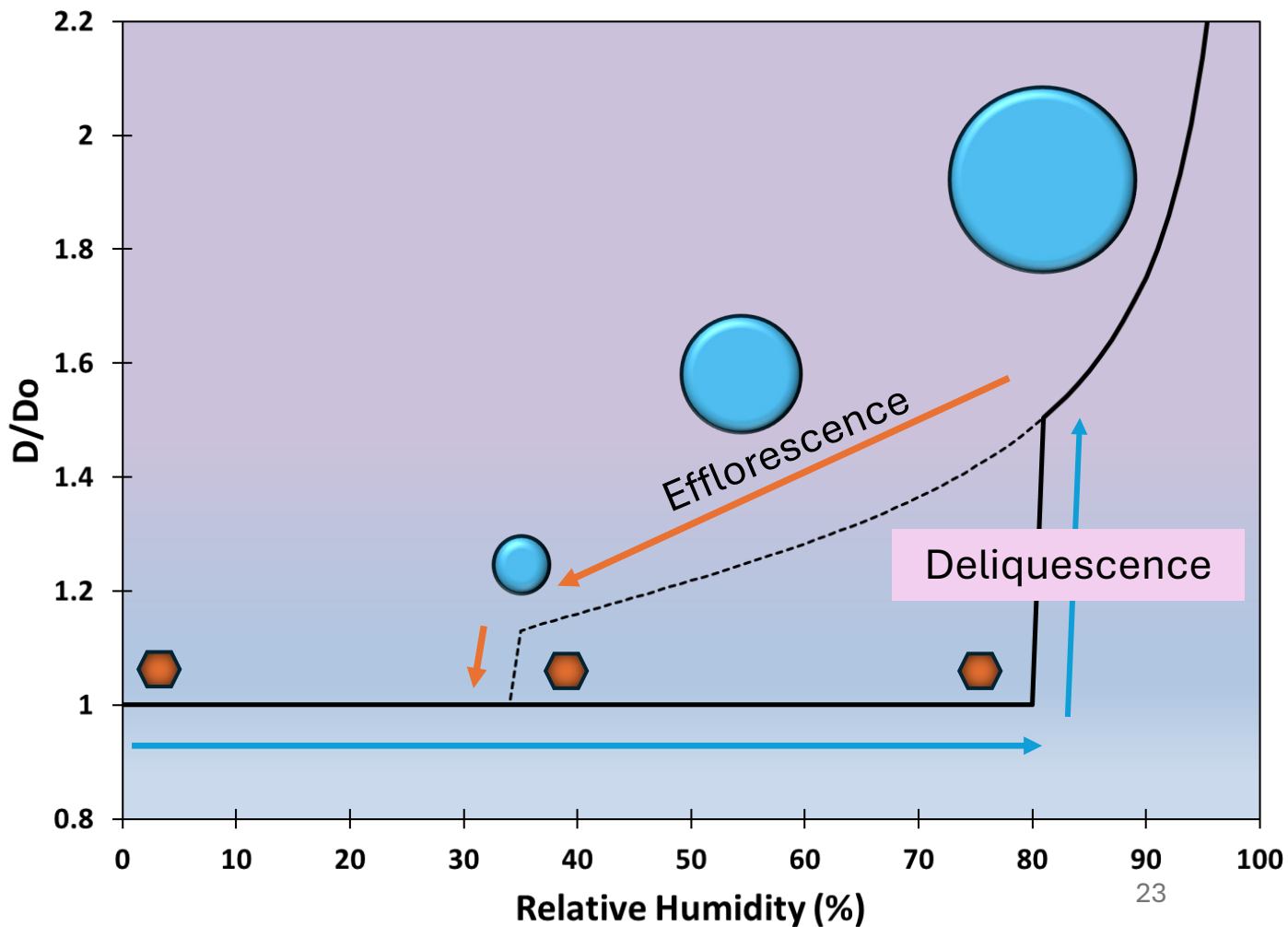
## Deliquescence



## Efflorescence

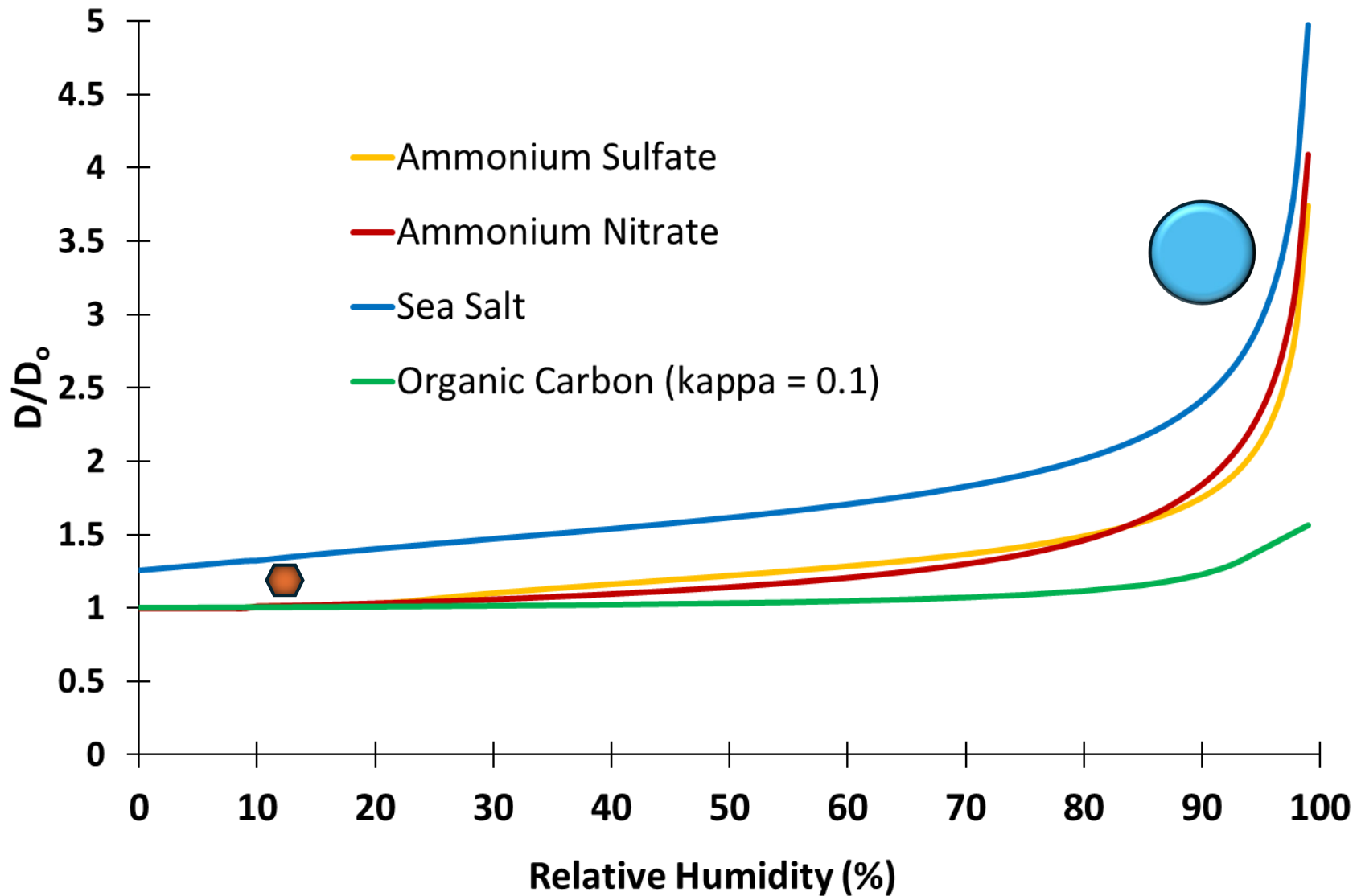


## Wet diameter/Dry diameter





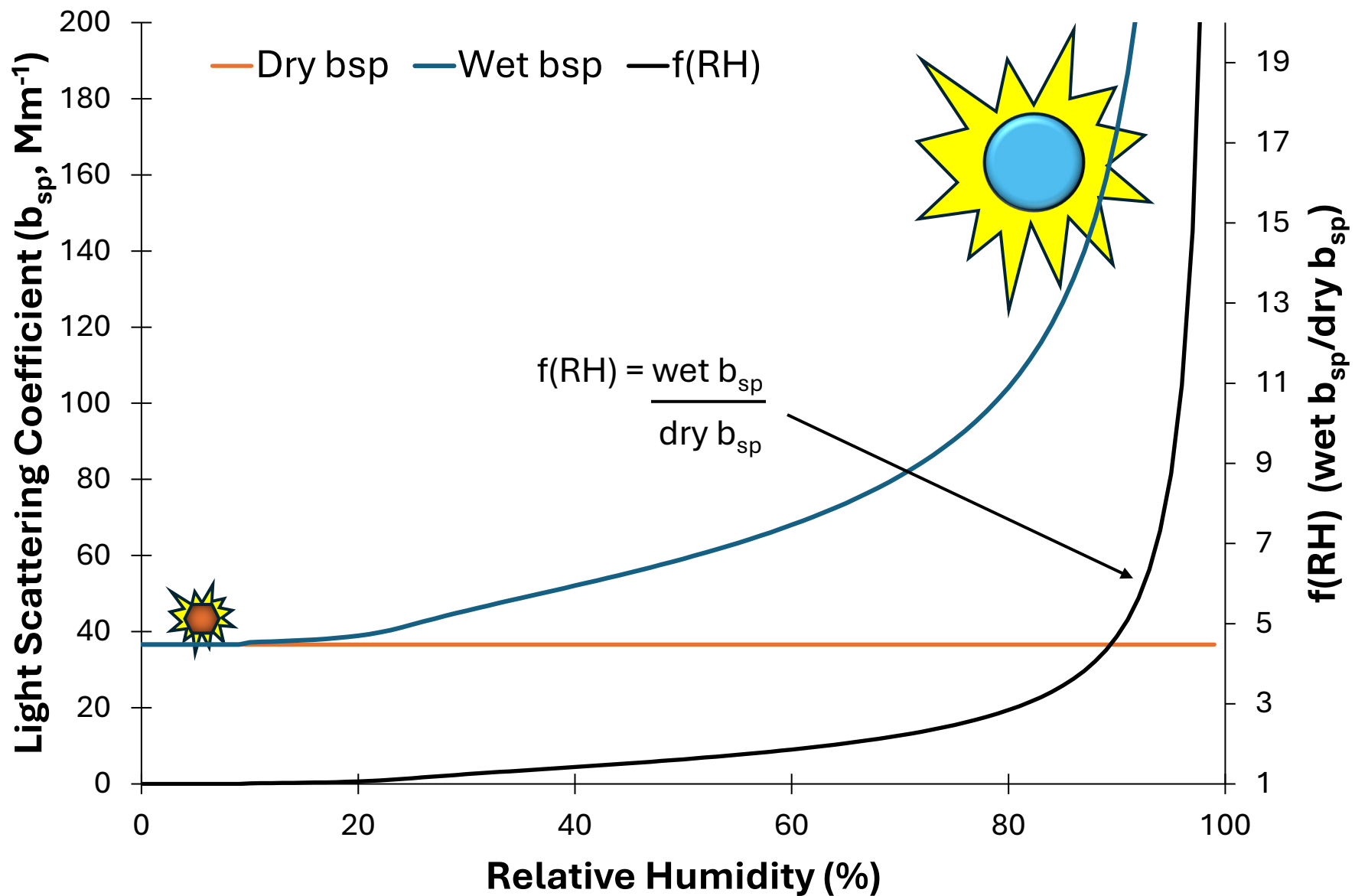
# Aerosol Diameter Growth Curves ( $D/D_0$ )







# Scattering Growth Curves (f(RH))





# Relative Humidity Effects

## Glacier National Park

**10  $\mu\text{g m}^{-3}$  Ammonium Sulfate**

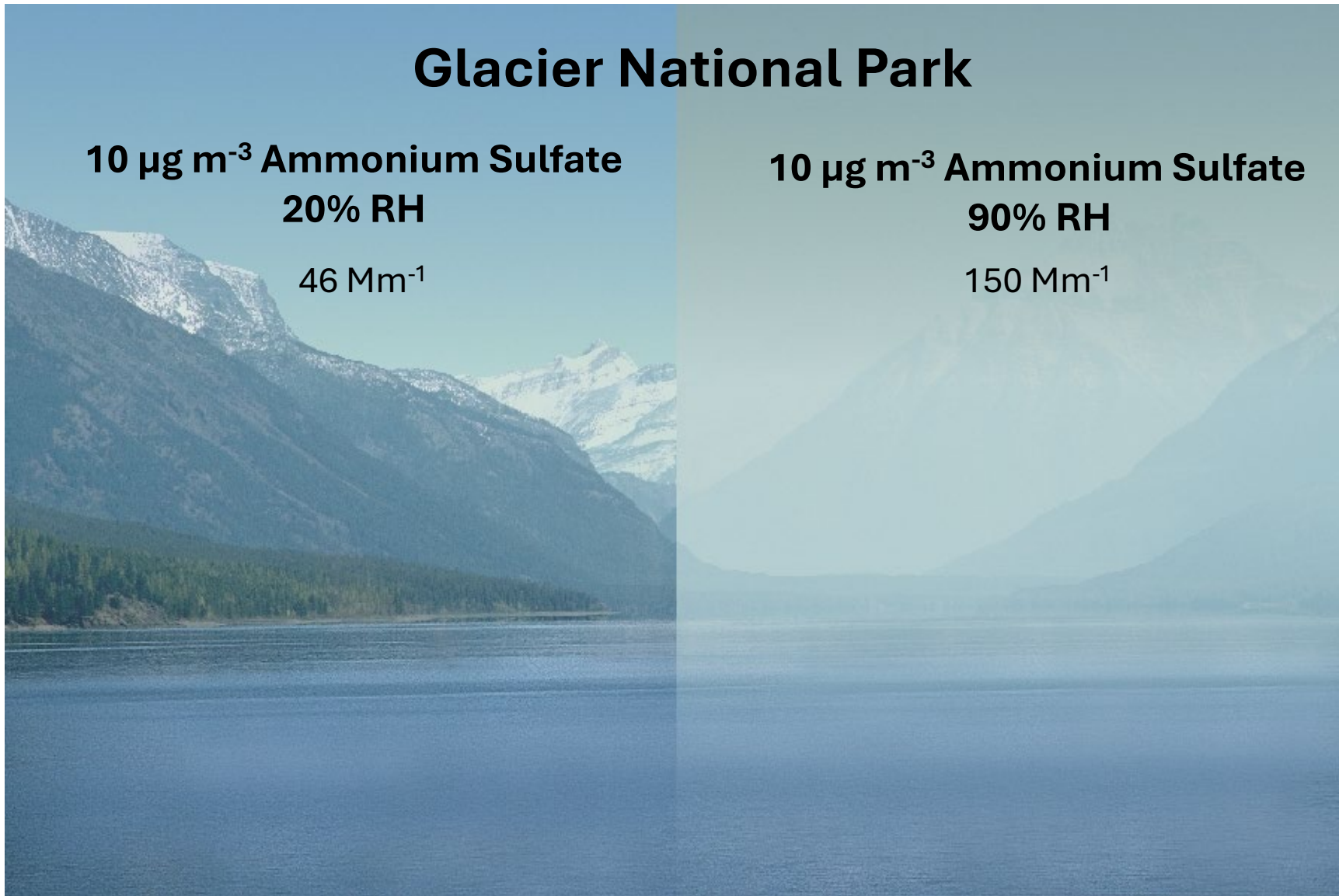
**20% RH**

**46  $\text{Mm}^{-1}$**

**10  $\mu\text{g m}^{-3}$  Ammonium Sulfate**

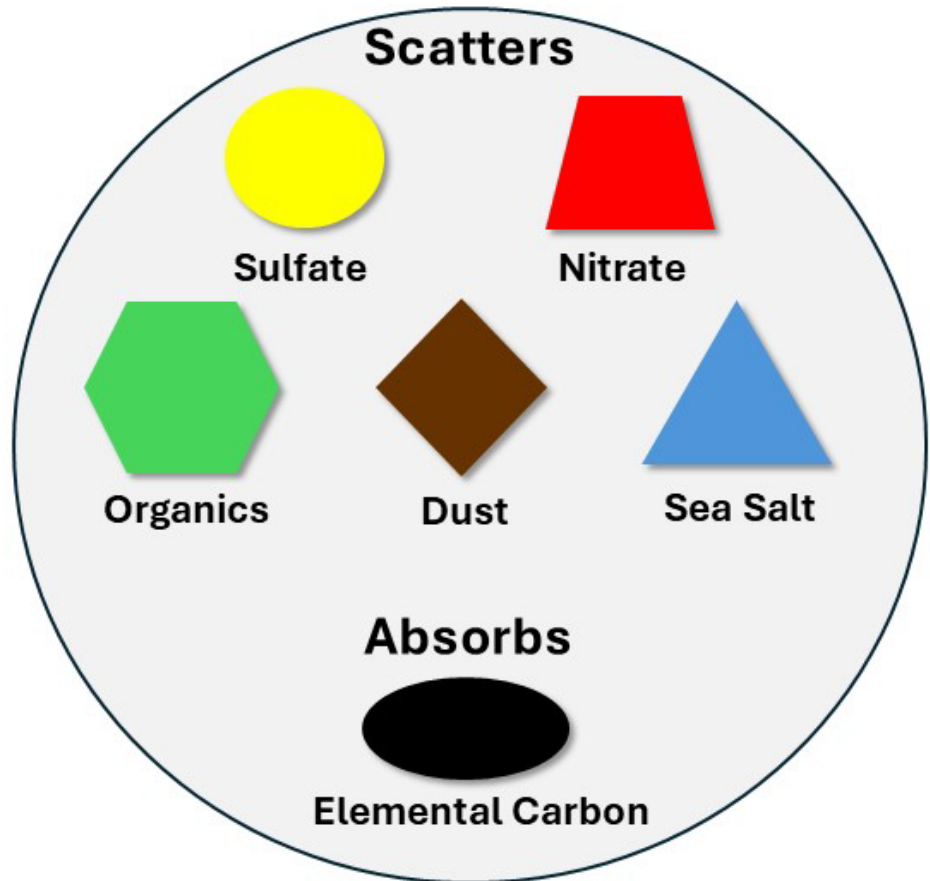
**90% RH**

**150  $\text{Mm}^{-1}$**





# Light Extinction Coefficient ( $b_{\text{ext}}$ )



Major Contributors

## 1<sup>st</sup> IMPROVE Equation

$$b_{\text{ext}} (\text{Mm}^{-1}) = 3.0 \times f(\text{rh}) \times [\text{Ammonium Sulfate}] + 3.0 \times f(\text{rh}) \times [\text{Ammonium Nitrate}] + 4.0 \times [\text{Organics}] + 1.0 \times [\text{Fine Dust}] + 0.6 \times [\text{Coarse Mass}] + 10.0 \times [\text{Elemental Carbon}] + \text{Rayleigh Scattering}$$

Dry mass scattering efficiency ( $\text{m}^2 \text{g}^{-1}$ )

Rayleigh Scattering of light by gas molecules:

Depends on the density of the air so is most dependent on elevation above sea level

Ranges from  $\sim 9\text{-}12 \text{ Mm}^{-1}$

$\lambda = 550 \text{ nm}$

Malm et al.<sup>27</sup> (1994)



# Dry Mass Scattering Efficiency ( $\alpha$ , $\text{m}^2\text{g}^{-1}$ )

## Light scattering coefficient ( $b_{\text{sp}}$ )

$$b_{\text{sp}} = \sum_i \frac{3}{2} \frac{M_i Q_{\text{sp},i}}{D_{p,i} \rho} \quad (\text{Mm}^{-1})$$

$M_i$  = mass of particles in bin  $i$

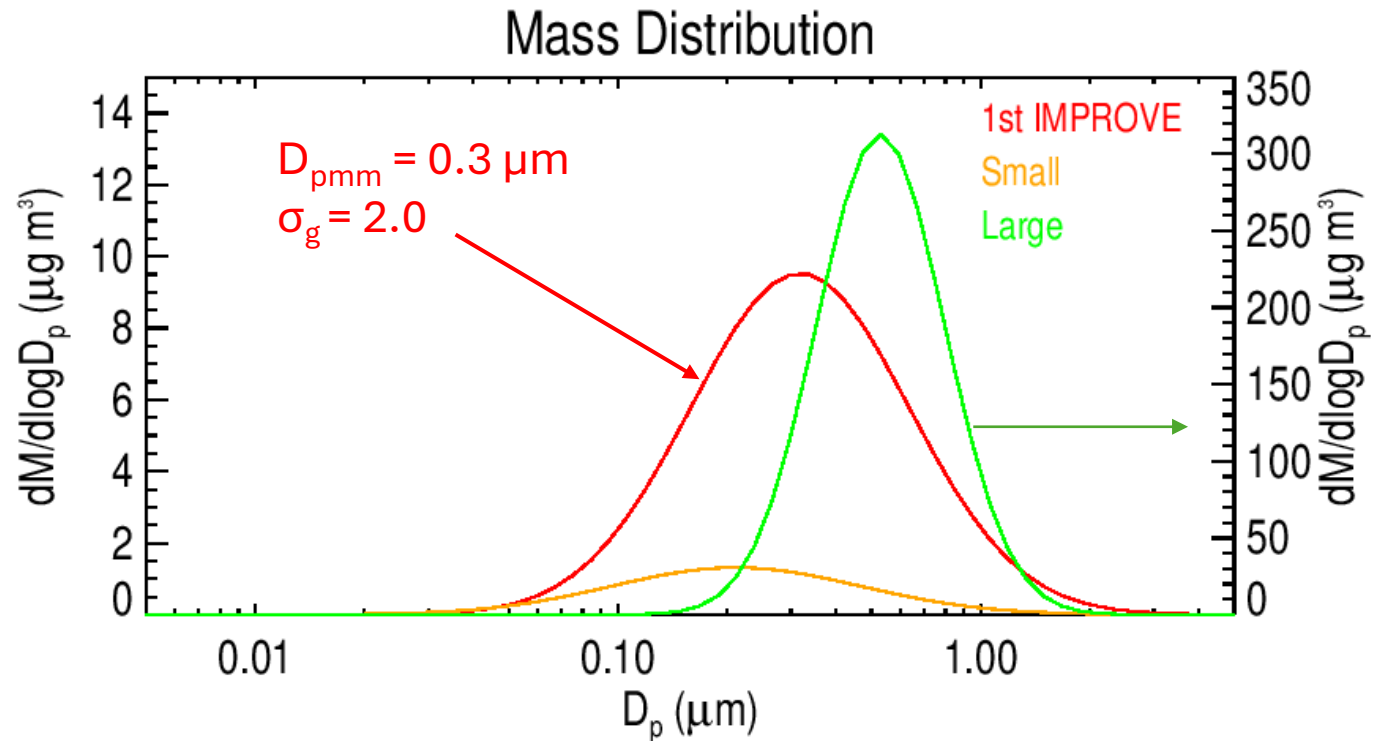
$D_{p,i}$  = diameter of bin  $i$

$Q_{\text{sp},i}$  = Mie scattering efficiency

$\rho$  = species density ( $\text{g cm}^{-3}$ )

## Mass scattering efficiency ( $\alpha$ )

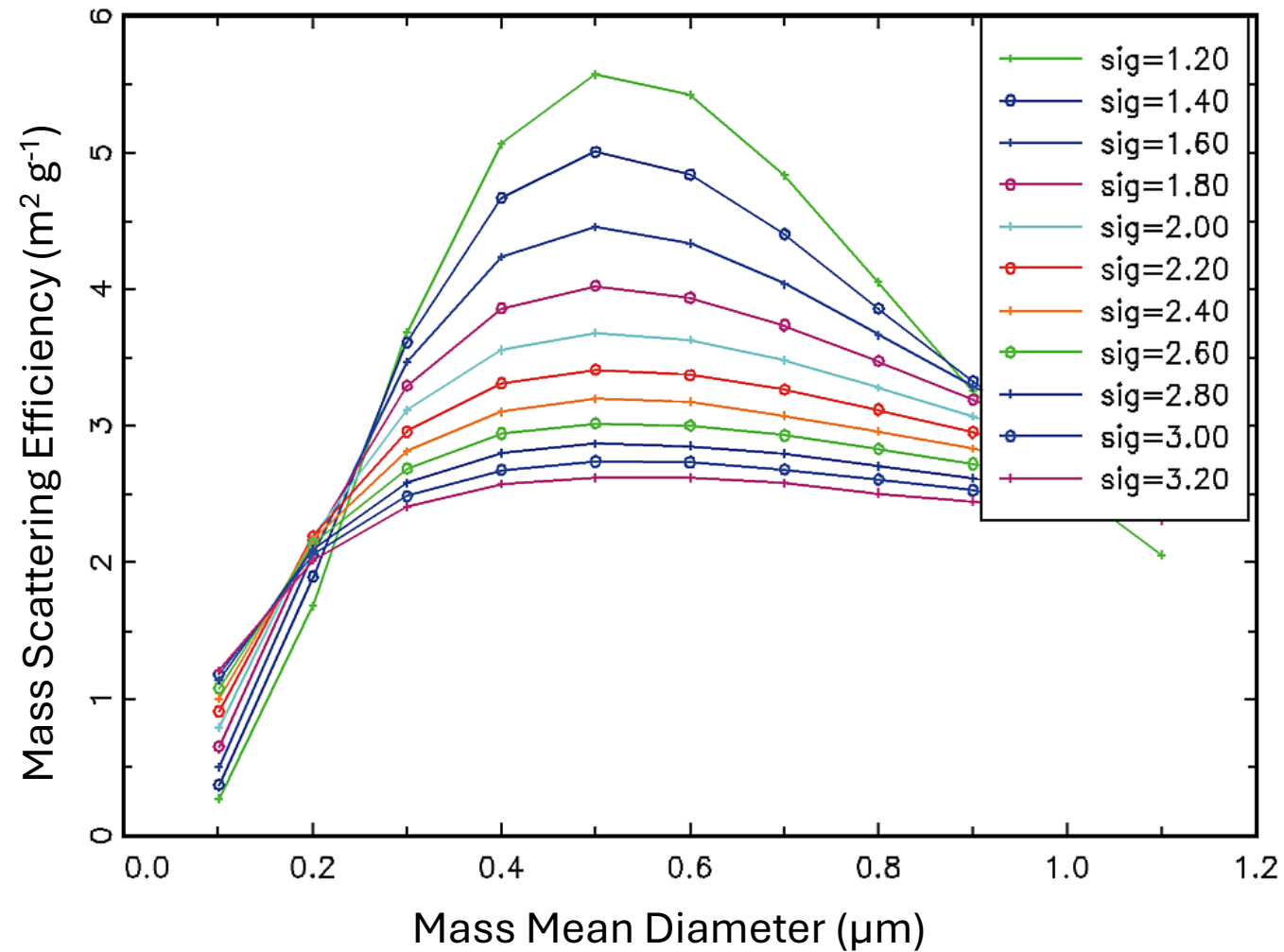
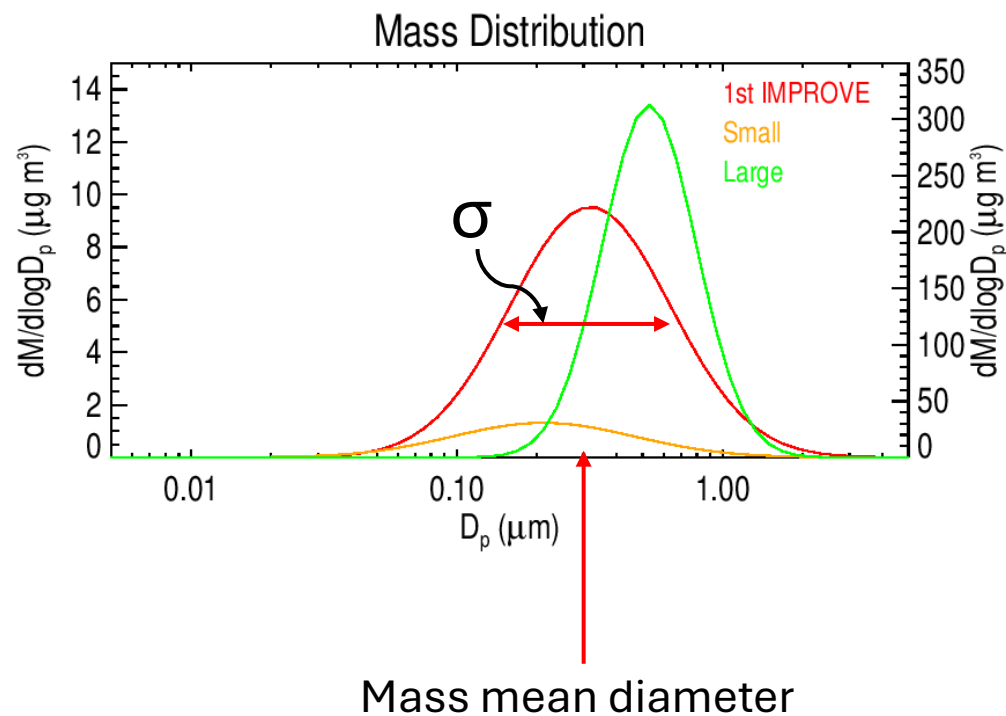
$$\alpha = b_{\text{sp}} / \text{Mass} \quad (\text{m}^2\text{g}^{-1})$$



Mie scattering efficiency ( $Q_{\text{sp}}$ ) is a function of refractive index, particle diameter, and wavelength, assuming spherical particles.

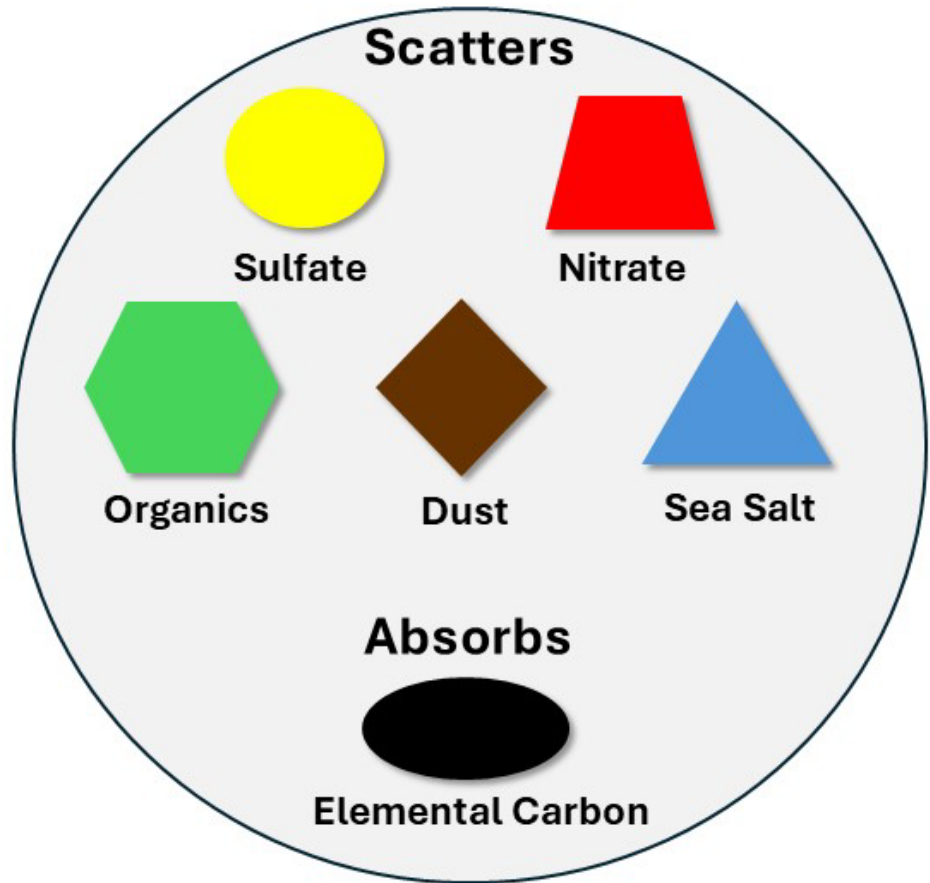


# Mass Scattering Efficiency as Function of Size





# Light Extinction Coefficient



Major Contributors

## 1<sup>st</sup> IMPROVE Equation

$$b_{\text{ext}} (\text{Mm}^{-1}) = 3.0 \times f(\text{RH}) \times [\text{Ammonium Sulfate}] + 3.0 \times f(\text{RH}) \times [\text{Ammonium Nitrate}] + 4.0 \times [\text{Organics}] + 1.0 \times [\text{Fine Dust}] + 0.6 \times [\text{Coarse Mass}] + 10.0 \times [\text{Elemental Carbon}] + \text{Rayleigh Scattering}$$

Scattering growth curves

# Calculating f(RH)

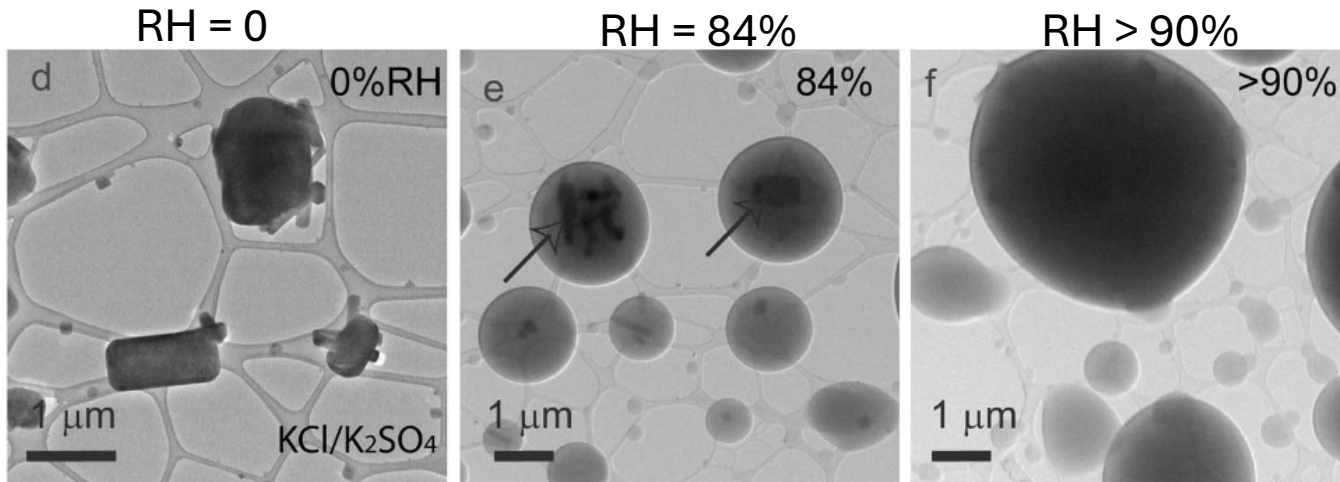
**DRY light scattering coefficient ( $b_{sp}$ )**

$$b_{sp} = \sum_i \frac{3}{2} \frac{M_i Q_{sp,i}}{D_{p,i} \rho} \quad (\text{Mm}^{-1})$$

**“Grow” the size distribution using  $D/D_0$  and calculate a WET  $b_{sp}$**

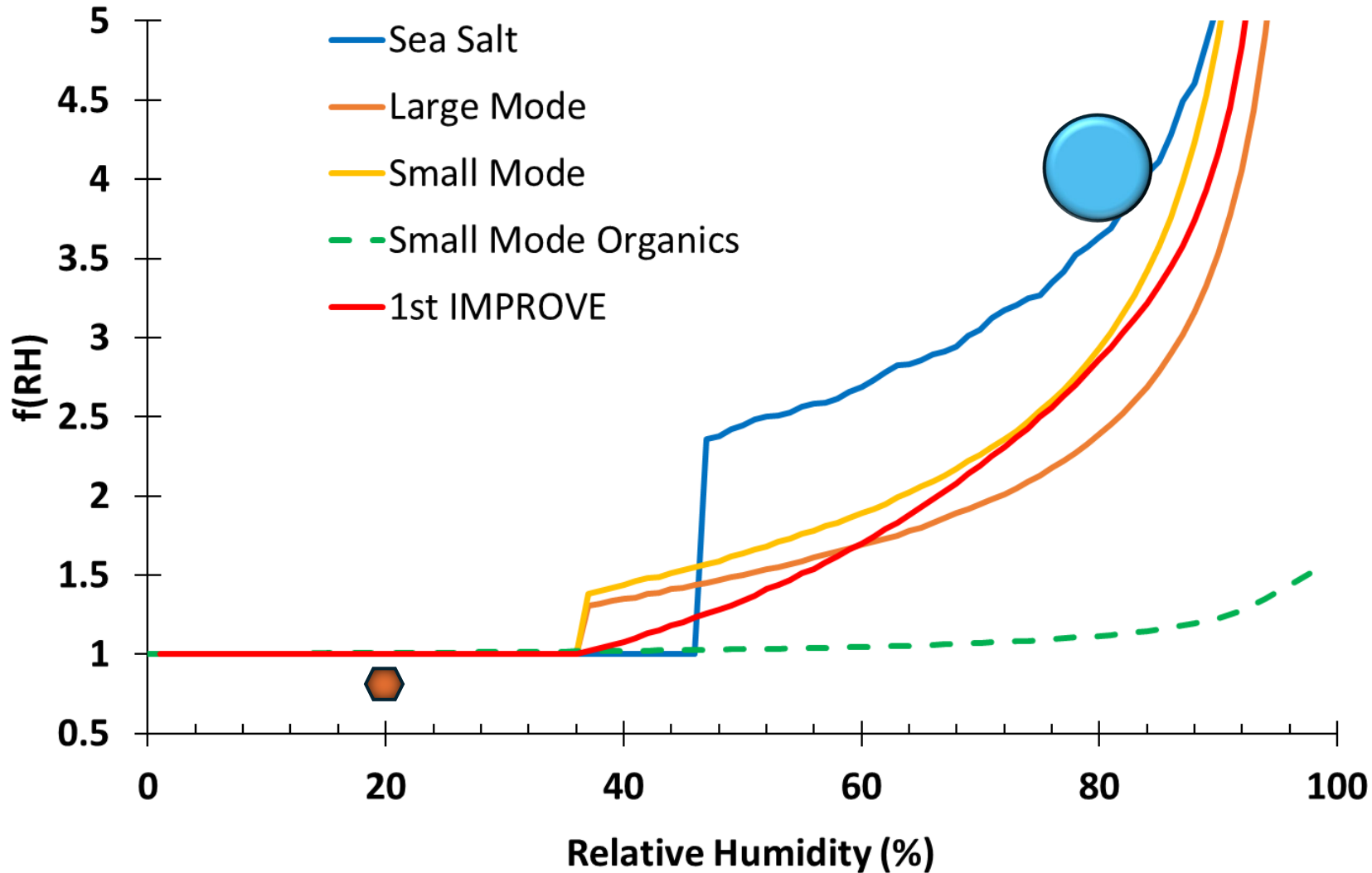


$$f(\text{RH}) = \frac{\text{wet } b_{sp}}{\text{dry } b_{sp}}$$





# Scattering Growth Curves (f(RH))







# 4. Visibility Measurements

## Optec NGN-2 Nephelometer



Photo courtesy of Mackenzie Reed/NPS

Optec NGN2: started in 1992  
Open Air nephelometer (total  $b_{sp}$ )  
Wavelength of 550 nm  
Ambient scattering without affecting RH of sample  
After 30 years, shut down on July 12, 2023



Photo courtesy of Mark Tigges, ARS



# New Nephelometers

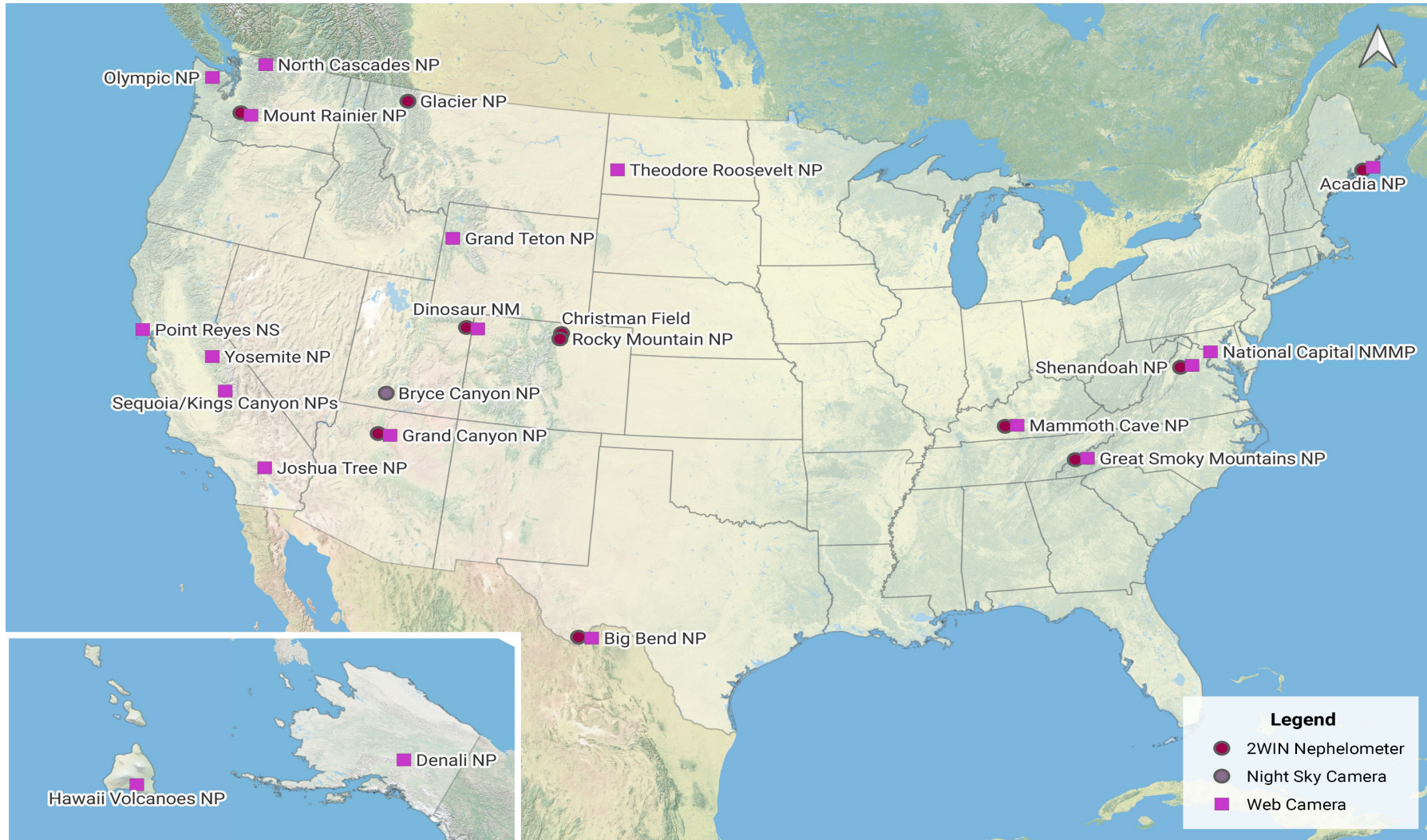
Ambilabs 2WIN Dual Wavelength nephelometer

Wavelength: 450nm, 525nm, and 635nm  
(only two at any given time)  
2.5  $\mu\text{m}$  size cut





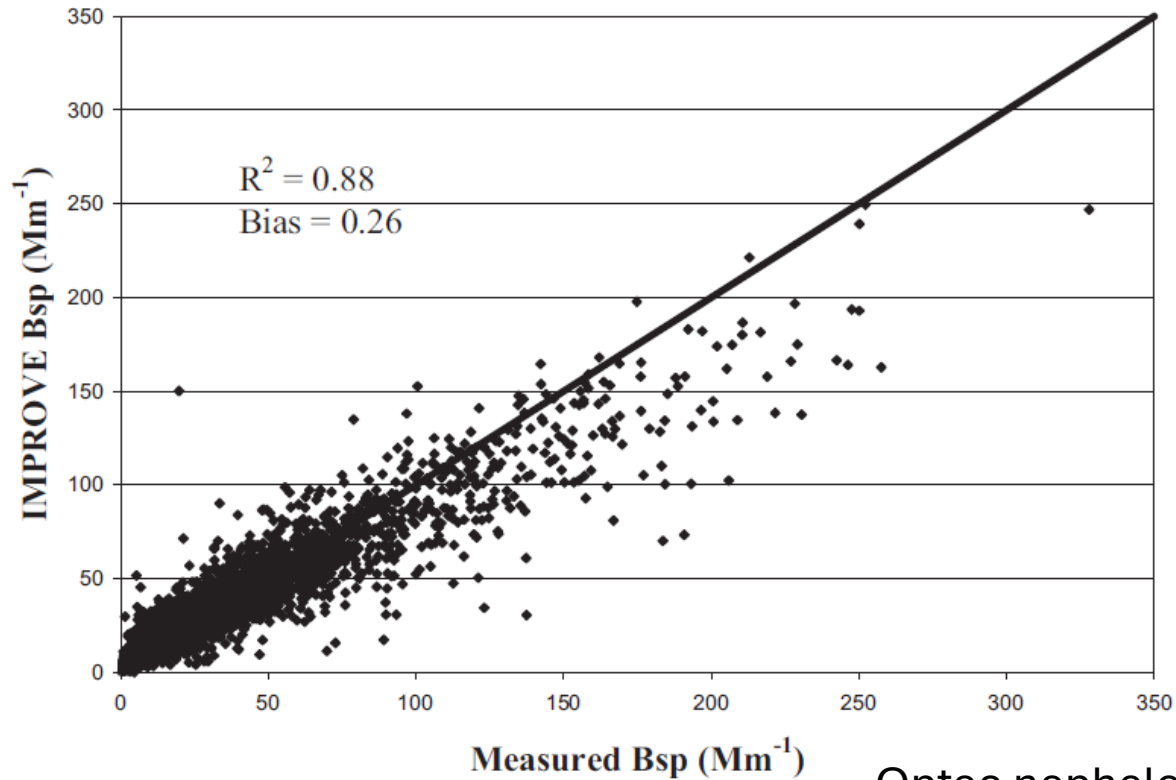
# New Nephelometers





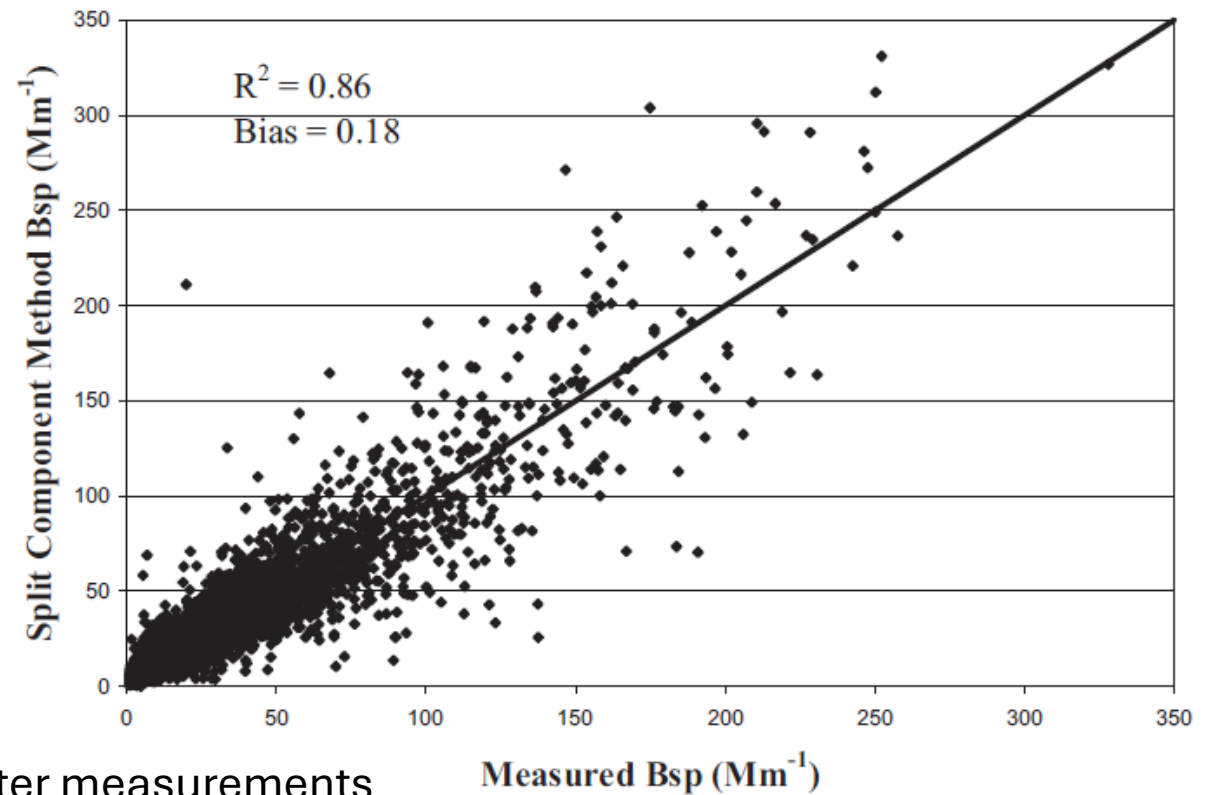
# Measured and Calculated Scattering

1<sup>st</sup> IMPROVE Equation (2001-2003)



Optec nephelometer measurements

2<sup>nd</sup> IMPROVE Equation



2<sup>nd</sup> IMPROVE Equation was designed to improve biases at low and high scattering values



## 5. 2<sup>nd</sup> IMPROVE Equation

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$$\begin{aligned} b_{\text{ext}} = & \mathbf{2.2 \times f_s(RH) \times [Small Ammonium Sulfate] +} \\ & \mathbf{4.8 \times f_L(RH) \times [Large Ammonium Sulfate] +} \\ & \mathbf{2.4 \times f_s(RH) \times [Small Ammonium Nitrate] +} \\ & \mathbf{5.1 \times f_L(RH) \times [Large Ammonium Nitrate] +} \\ & \mathbf{2.8 \times [Small Organic Mass] +} \\ & \mathbf{6.1 \times [Large Organic Mass] +} \\ & \mathbf{1 \times [Fine Dust] +} \\ & \mathbf{1.7 \times f_{ss}(RH) \times [Sea Salt] +} \\ & \mathbf{0.6 \times [Coarse Mass] +} \\ & \mathbf{10 \times [Elemental Carbon] +} \\ & \mathbf{Rayleigh Scattering (site specific)} \end{aligned}$$

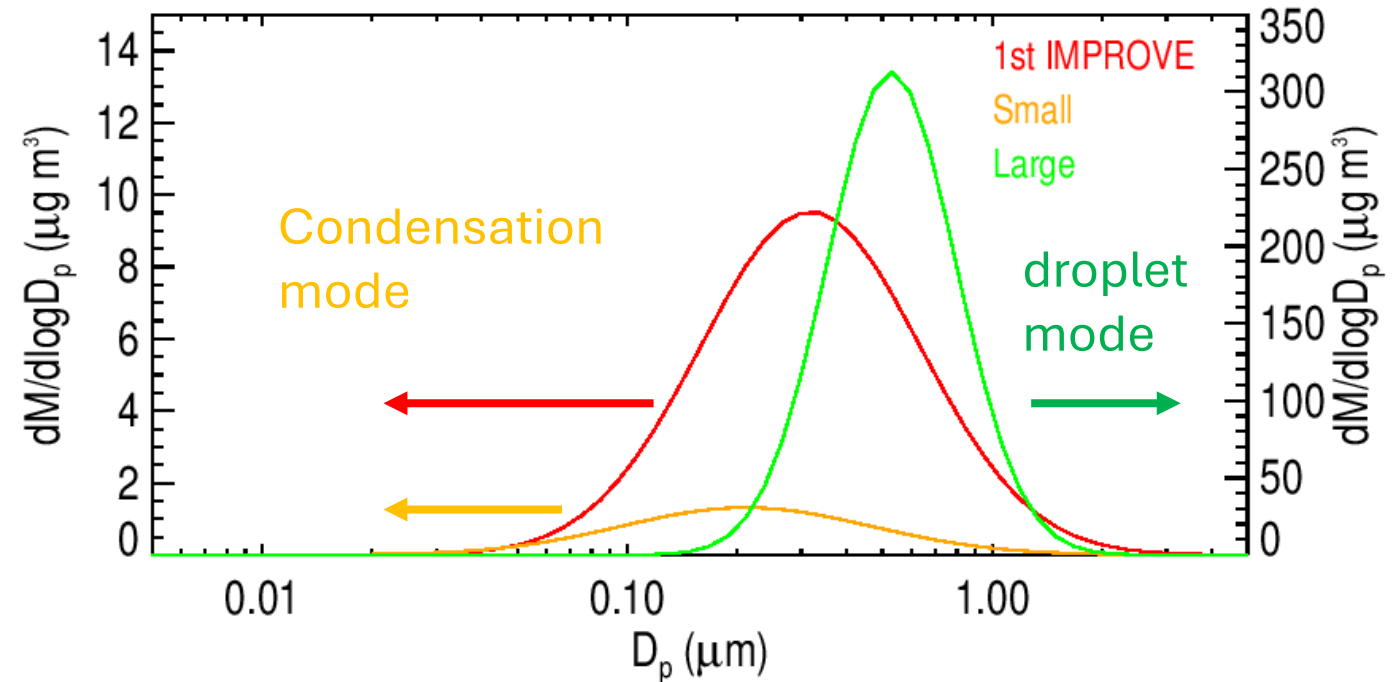


# 2<sup>nd</sup> IMPROVE Equation

The **2<sup>nd</sup> IMPROVE Equation** is also known as the “**split mode**” or “**split component**” algorithm because it splits, or apportions, the component PM<sub>2.5</sub> mass into two size modes.

Mode	D <sub>pmm</sub> (μm)	σ <sub>g</sub>
1 <sup>st</sup> IMPROVE	0.3	2.0
2 <sup>nd</sup> IMPROVE (small)	0.2	2.2
2 <sup>nd</sup> IMPROVE (large)	0.5	1.5
2 <sup>nd</sup> IMPROVE (sea salt)	2.5	2.0

Mass Distribution





# Split Component Algorithm

The mass in the two modes is apportioned using a “cut point” of  $20 \mu\text{g m}^{-3}$ .

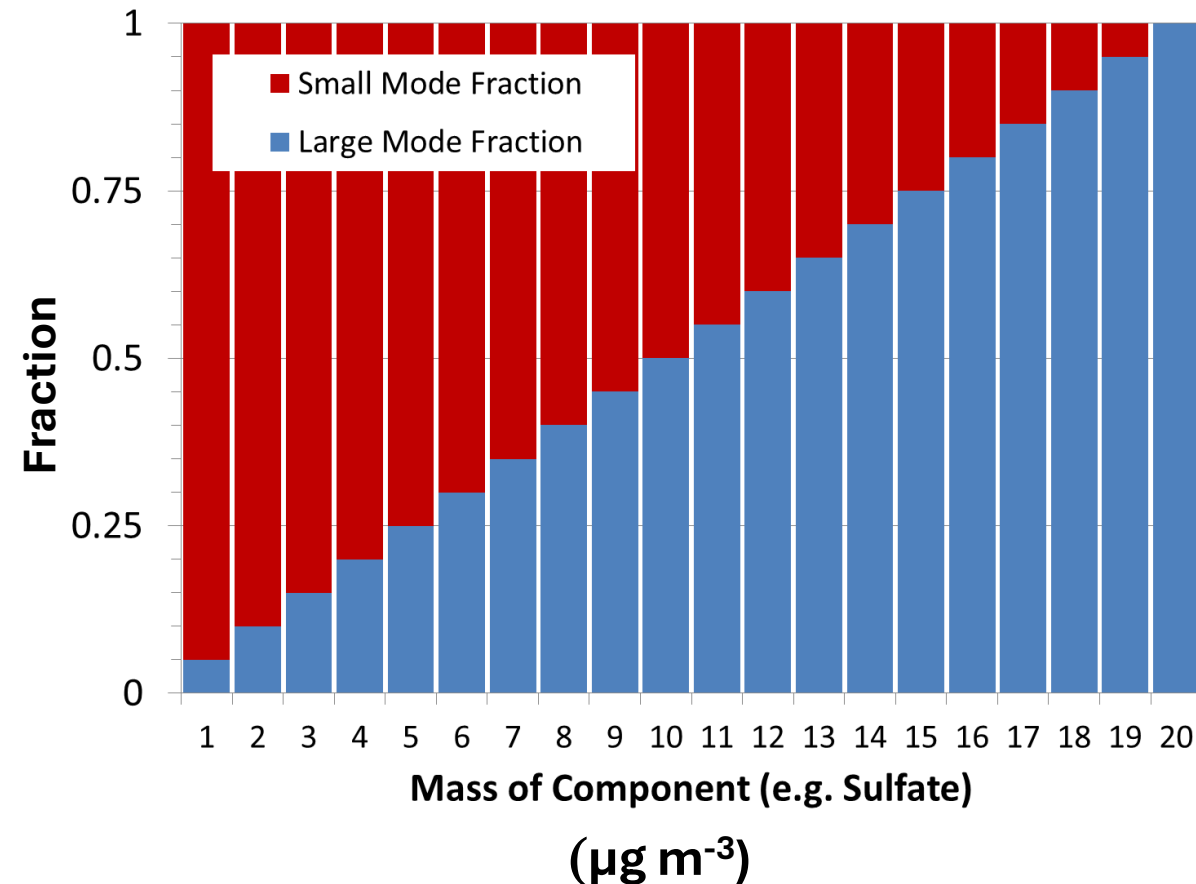
For example:

If the  $\text{PM}_{2.5}$  ammonium sulfate mass (AS) is  $4 \mu\text{g m}^{-3}$ ,

- “large mode” AS is one fifth ( $4/20$ ) of the mass, or  $0.8 \mu\text{g m}^{-3}$
- “small mode” AS mass is  $4 - 0.8 = 3.2 \mu\text{g m}^{-3}$ .

If the AS more than  $20 \mu\text{g m}^{-3}$ , all of it is assumed to be in the “large mode”.

Ammonium nitrate and organic mass are also apportioned using  $20 \mu\text{g m}^{-3}$ .





# Second IMPROVE Equation

Dry mass  
scattering  
efficiency ( $\text{m}^2 \text{g}^{-1}$ )

$$b_{\text{ext}} = 2.2 \times f_s(\text{RH}) \times [\textit{Small Sulfate}] +$$
$$4.8 \times f_L(\text{RH}) \times [\textit{Large Sulfate}] +$$
$$2.4 \times f_s(\text{RH}) \times [\textit{Small Nitrate}] +$$
$$5.1 \times f_L(\text{RH}) \times [\textit{Large Sulfate}] +$$
$$2.8 \times [\textit{Small Organic Mass}] +$$
$$6.1 \times [\textit{Large Organic Mass}] +$$
$$1 \times [\textit{Fine Soil}] +$$
$$1.7 \times f_{\text{ss}}(\text{RH}) \times [\textit{Sea Salt}] +$$
$$0.6 \times [\textit{Coarse Mass}] +$$
$$10 \times [\textit{Elemental Carbon}] +$$

Rayleigh Scattering (site specific)





# Split Component Mass Scattering Efficiencies

$$b_{sp} = \sum_i \frac{3 M_i Q_{sp,i}}{2 D_{p,i} \rho} \quad (\text{Mm}^{-1})$$

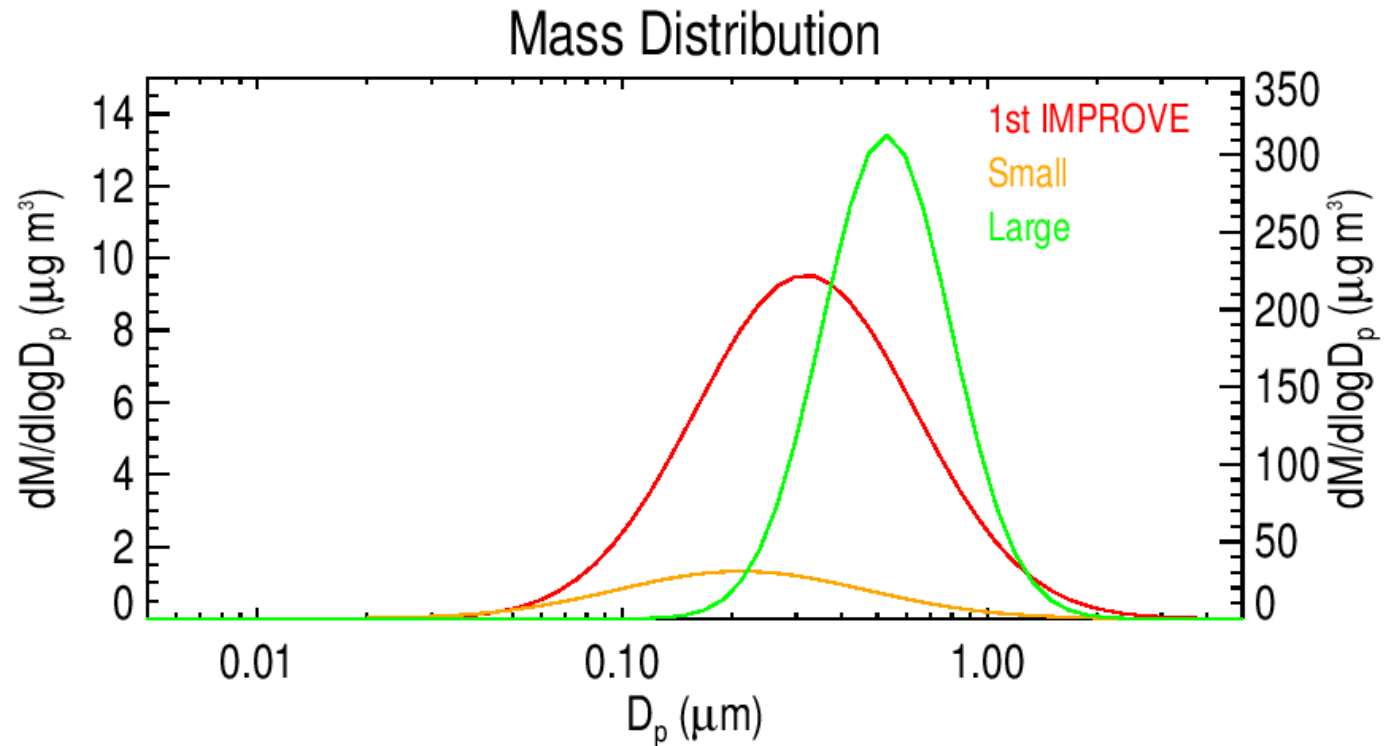
$M_i$  = mass of particles in bin  $i$

$D_{p,i}$  = diameter of bin  $i$

$Q_{sp,i}$  = Mie scattering efficiency

$\rho$  = density ( $\text{g cm}^{-3}$ )

$$\alpha = b_{sp} / \text{Mass} \quad (\text{m}^2 \text{g}^{-1})$$

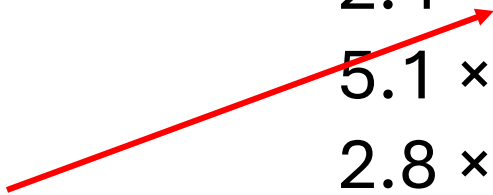




# Split Component f(RH)

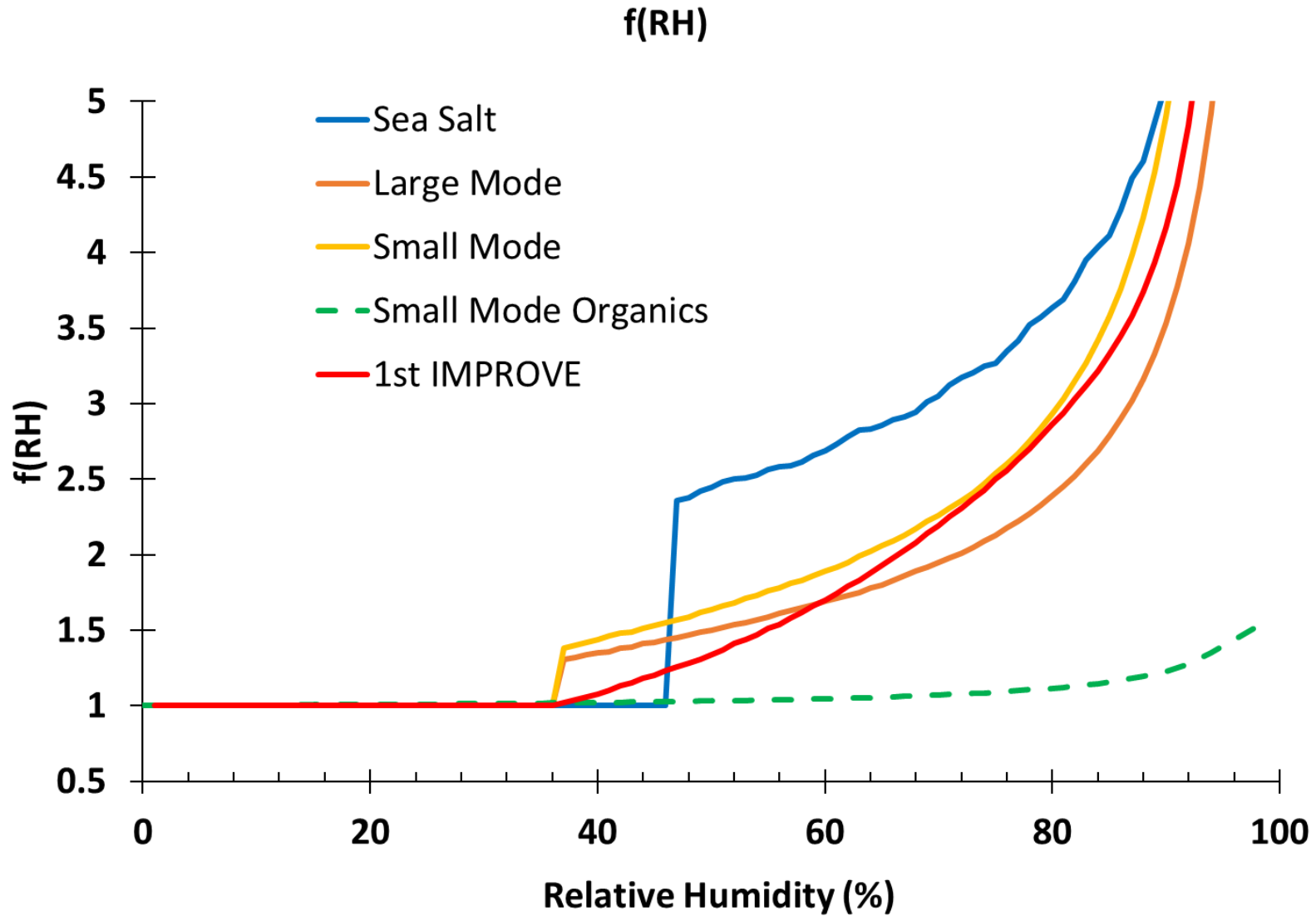
$$\begin{aligned} b_{\text{ext}} = & 2.2 \times f_s(\text{RH}) \times [\textit{Small Sulfate}] + \\ & 4.8 \times f_L(\text{RH}) \times [\textit{Large Sulfate}] + \\ & 2.4 \times f_s(\text{RH}) \times [\textit{Small Nitrate}] + \\ & 5.1 \times f_L(\text{RH}) \times [\textit{Large Nitrate}] + \\ & 2.8 \times [\textit{Small Organic Mass}] + \\ & 6.1 \times [\textit{Large Organic Mass}] + \\ & 1 \times [\textit{Fine Soil}] + \\ & 1.7 \times f_{ss}(\text{RH}) \times [\textit{Sea Salt}] + \\ & 0.6 \times [\textit{Coarse Mass}] + \\ & 10 \times [\textit{Elemental Carbon}] + \\ & \text{Rayleigh Scattering (site specific)} \end{aligned}$$

Scattering  
growth curves





# Split Component $f(\text{RH})$

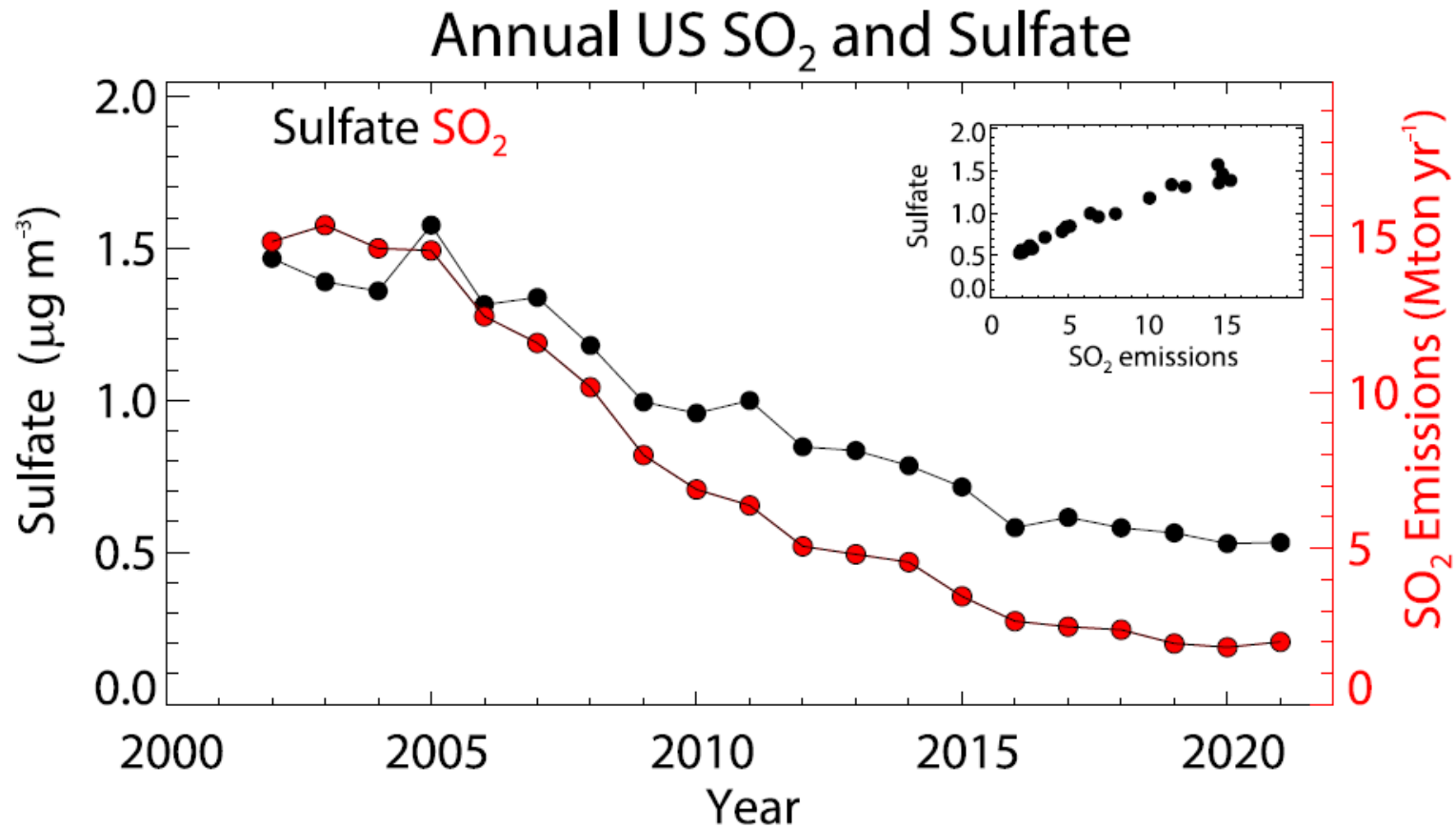




# 2<sup>nd</sup> IMPROVE Equation

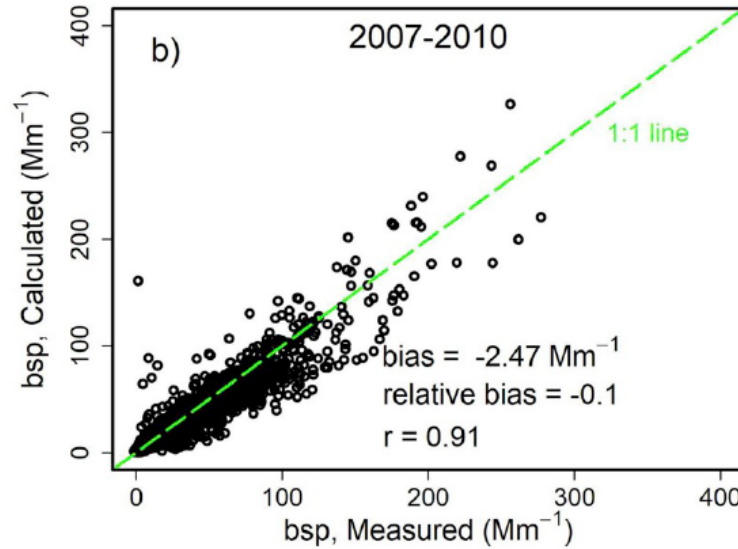
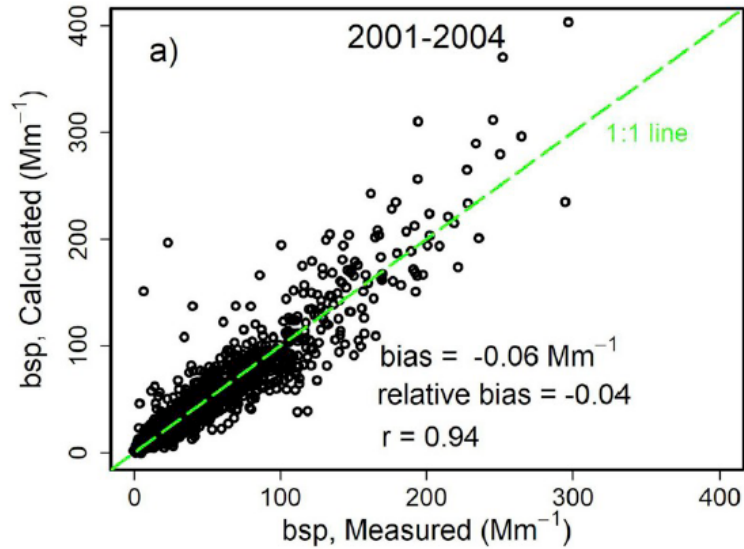
The “cut point” of  $20 \mu\text{g m}^{-3}$  is empirically derived based on IMPROVE data from 2001-2003.

The atmosphere has changed significantly since then.

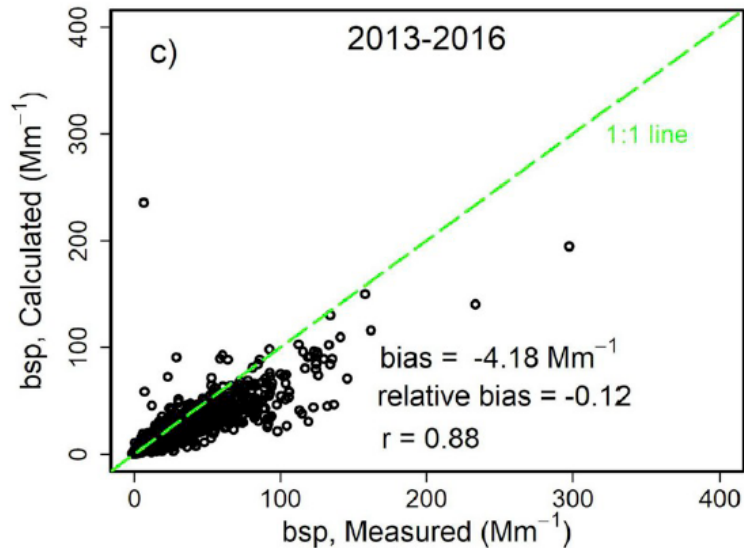




# Measured and Calculated Scattering



Optec nephelometer data & 2<sup>nd</sup> IMPROVE Equation (11 sites)



In recent years, we are underestimating reconstructed  $b_{\text{ext}}$  due to the split component aspect of the 2<sup>nd</sup> IMPROVE Equation.

# 6. Other Visibility Parameters

- Visual Range (VR)
- Deciview (dv)

## Rayleigh Conditions

$$b_{\text{ext}} = 9 \text{ Mm}^{-1}$$

$$dv = 0$$

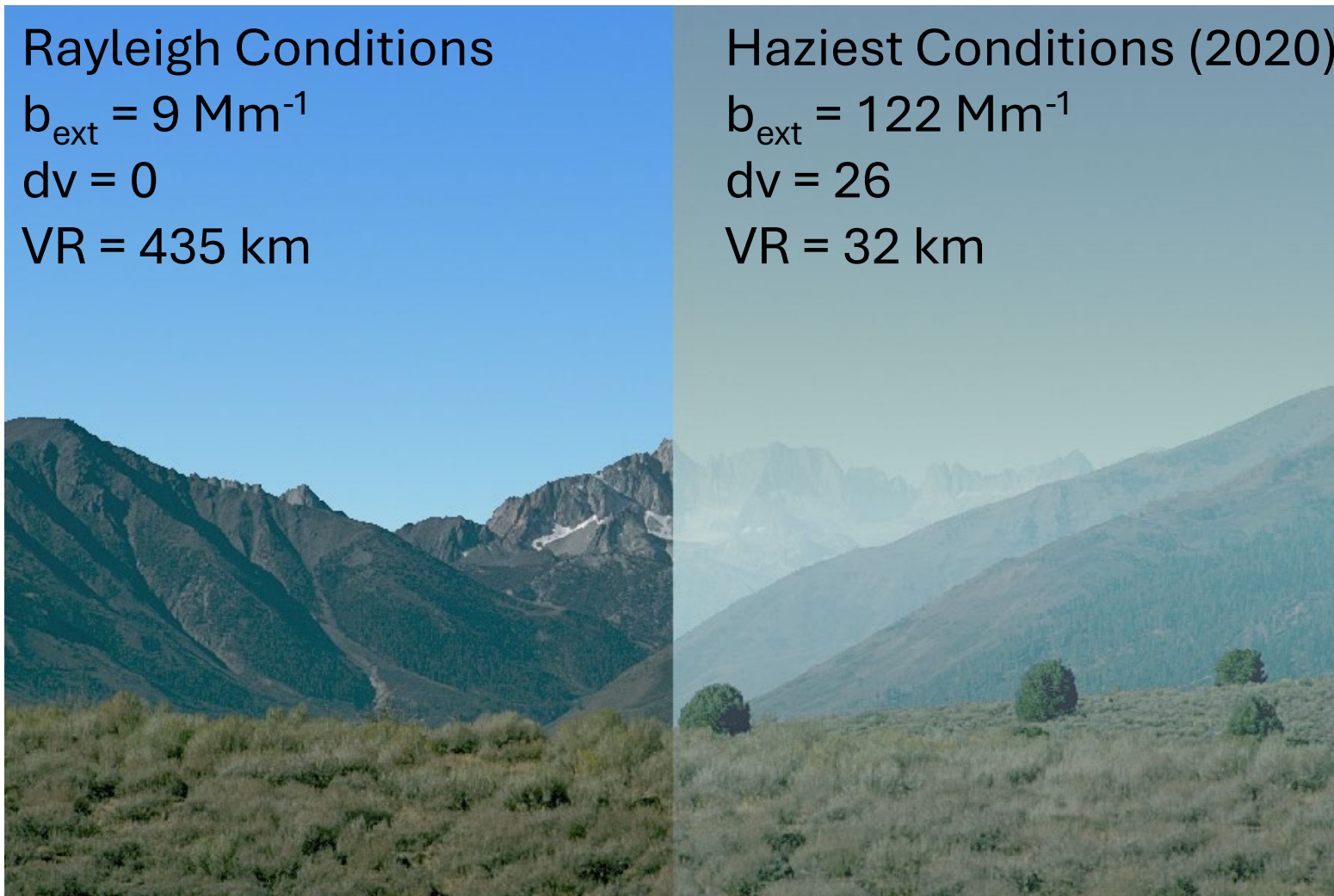
$$\text{VR} = 435 \text{ km}$$

## Haziest Conditions (2020)

$$b_{\text{ext}} = 122 \text{ Mm}^{-1}$$

$$dv = 26$$

$$\text{VR} = 32 \text{ km}$$



Hoover, CA



# Visual Range

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- The greatest distance that large dark objects can be seen.
- Visual range (VR) is the extinction-based index designed to estimate visual range.
- VR is defined by Koschmieder equation (VR in km and  $b_{\text{ext}}$  in  $\text{Mm}^{-1}$ ):

$$\mathbf{VR = 3910/b_{\text{ext}}}$$

- Advantage: Useful for when only the distance is important (aviation)
- Disadvantage: Doesn't account for loss of texture and discoloration with haze



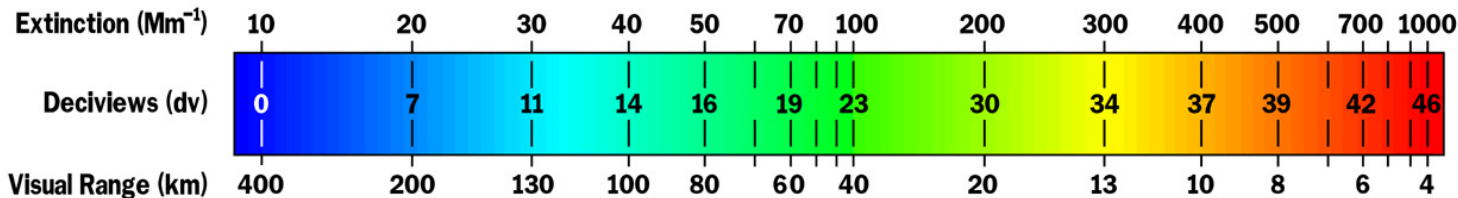
# Deciview (dv)

- Deciview is a haze index that expresses changes in scene quality.

$$\text{Haze Index (dv)} = 10 \ln(b_{\text{ext}}/10)$$

where  $b_{\text{ext}}$  is in  $\text{Mm}^{-1}$

- Linear with perceived changes in visibility.
- Haze index in deciview units is analogous to sound measured in decibel units in that they are both logarithmic transformation to produce perceptually linear parameters.





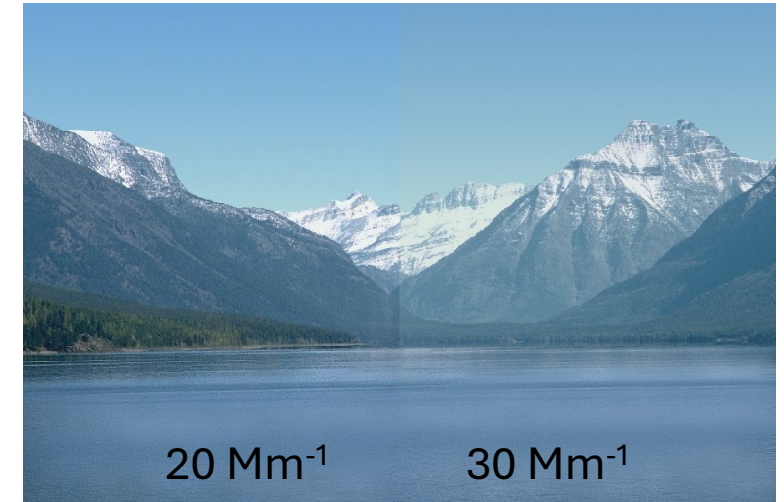
# Deciview

- $b_{\text{ext}}$  is non-linearly related to a person's perception of changes in haze.
  - $10 \text{ Mm}^{-1}$  increase in  $b_{\text{ext}}$  will have a larger perceived impact on a scene at  $b_{\text{ext}}$  of  $20 \text{ Mm}^{-1}$  than at  $b_{\text{ext}} = 100 \text{ Mm}^{-1}$ .

Advantage: under many circumstances a 1 dv change will be perceived the same on clear and hazy days

Disadvantage: not easily related to gas and aerosol concentrations

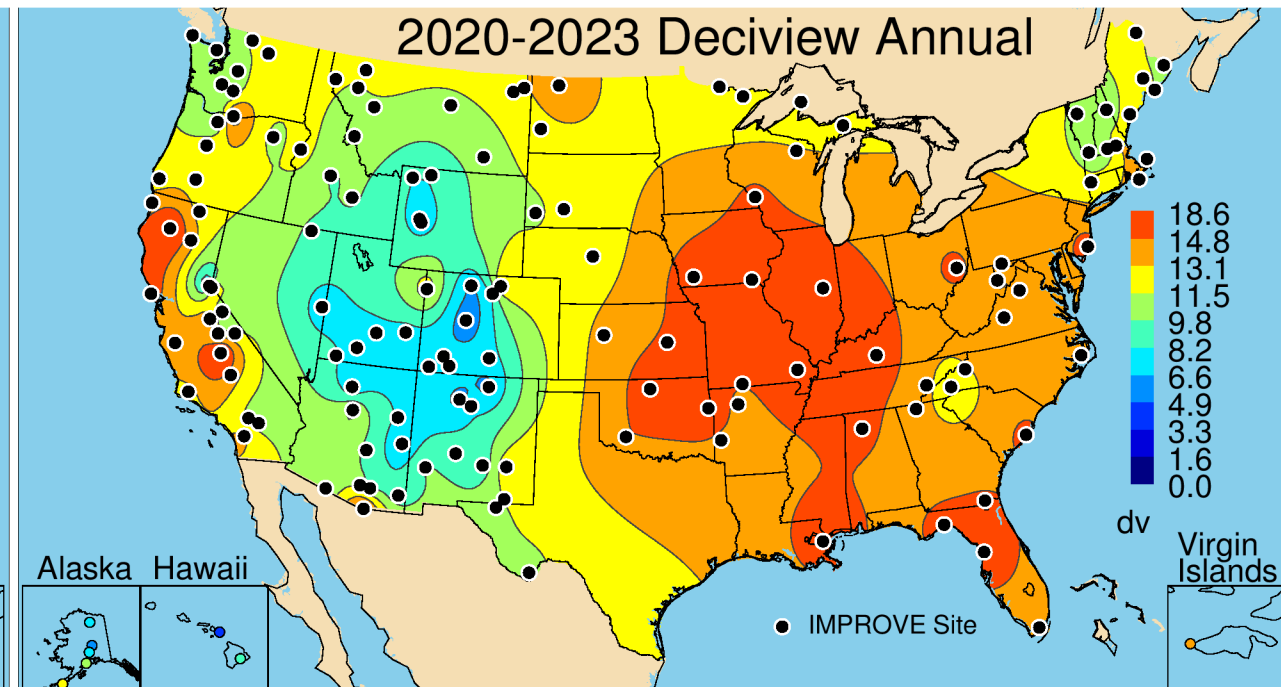
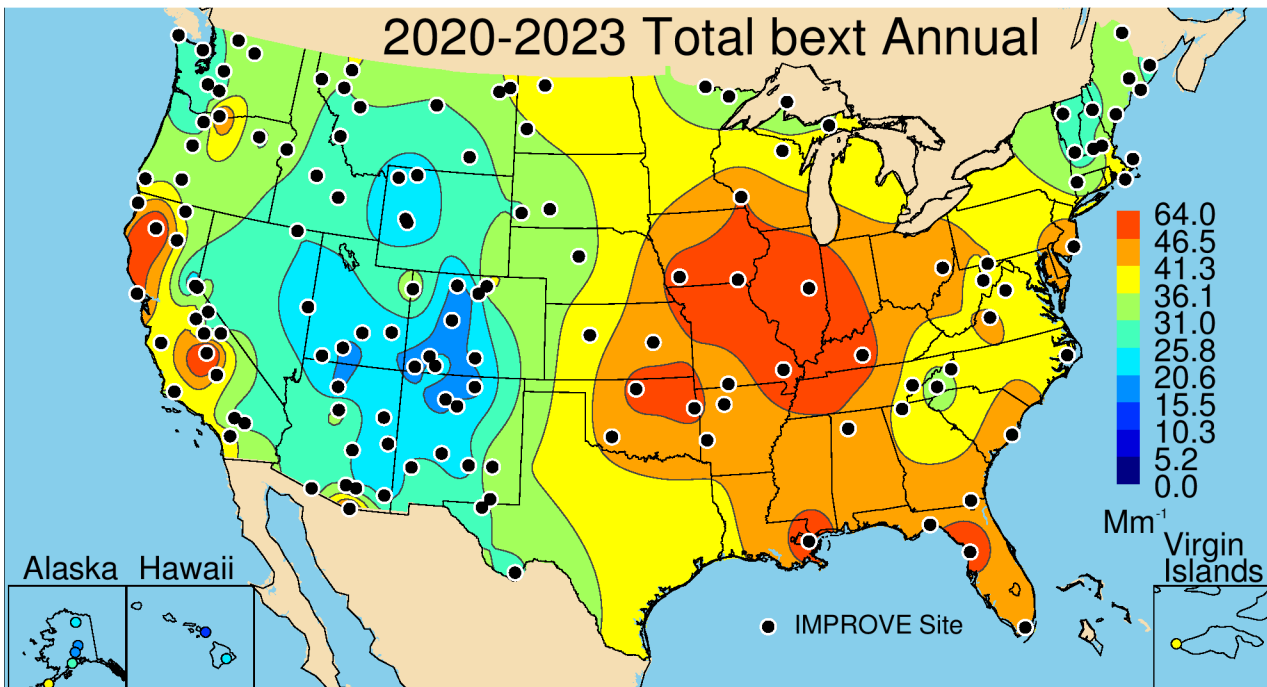
## Glacier National Park



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**Modeled with Winhaze**



# 2020-2023 Annual Mean Rural U.S. Visibility

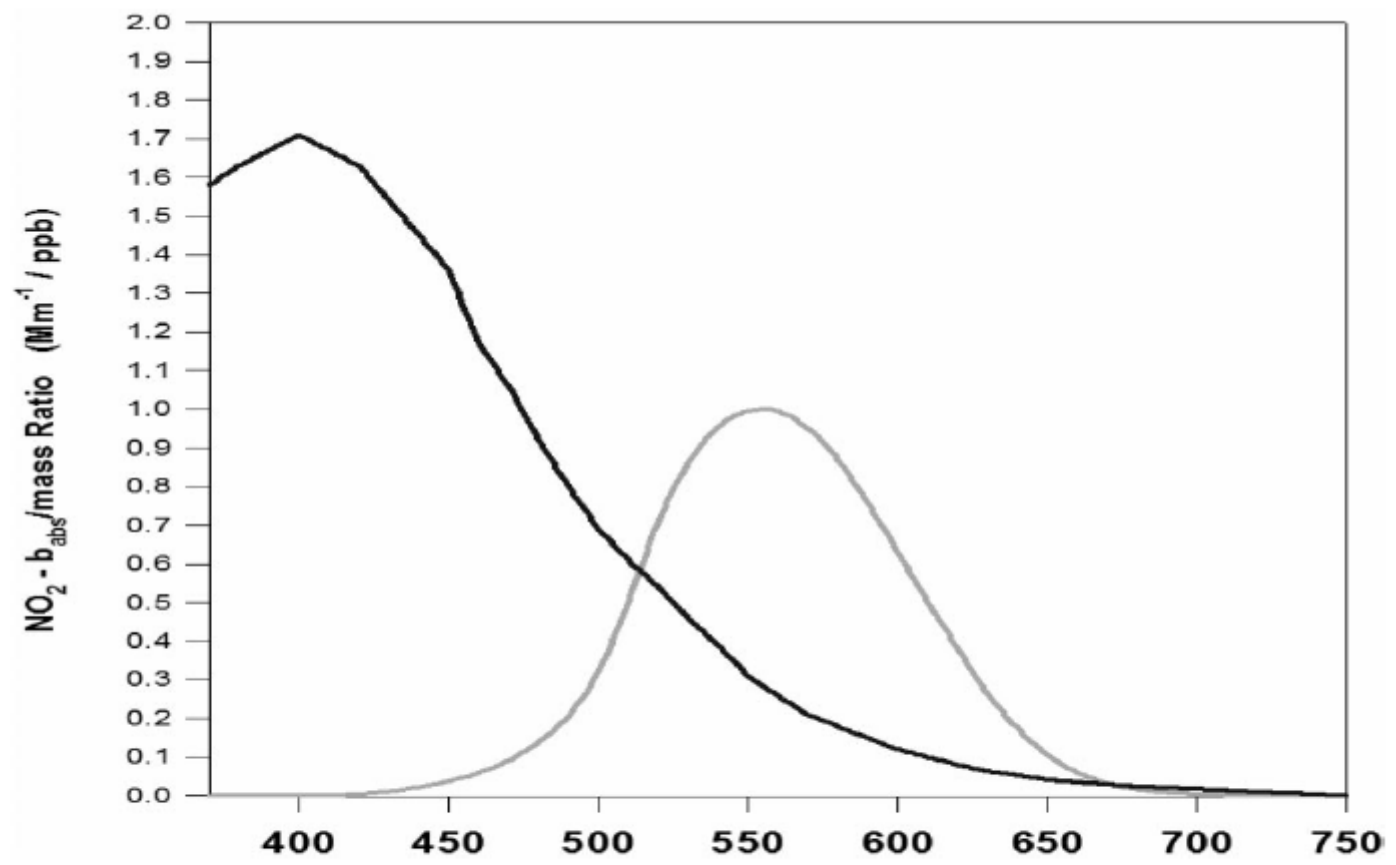




# Thanks!



Photo: Scott  
Copeland



**Figure 2.**  $\text{NO}_2$  absorption coefficient and 2° observer PR curves as a function of wavelength ( $\eta\text{m}$ ).



# Previous NPS Optical Monitoring Network

